

AMINO ACIDS CHEMISTRY

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Amino acids chemistry

LD-1

Chemistry and Biology. Chirality. α -world

Secondary structures of α -peptides

Chemistry: Protecting groups

Chemistry: Peptide Coupling. Solid-phase synthesis

LD-2

Unnatural world

β -amino acids and higher homologues of α -amino acids

Secondary structures of β and γ -peptides

Homologation of α -amino acids

LD-3

Unnatural world

Sulfur and Selenium containing unnatural amino acids

Proline based unnatural amino acids

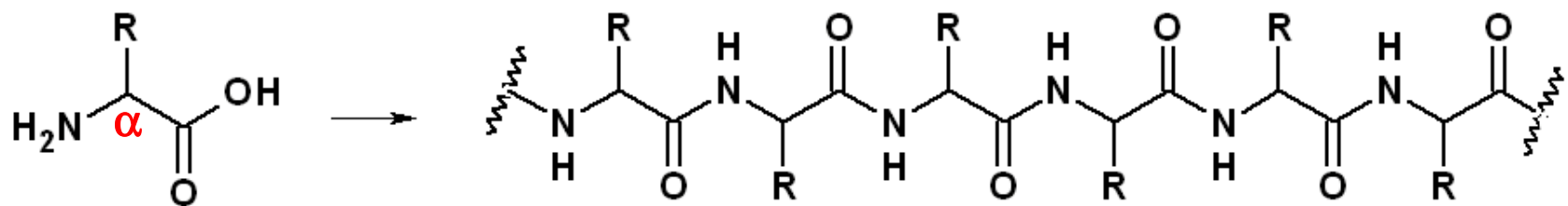
Cystathionine and Homocysteine families

LD-4

Peptidomimetics Chemistry

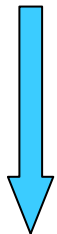
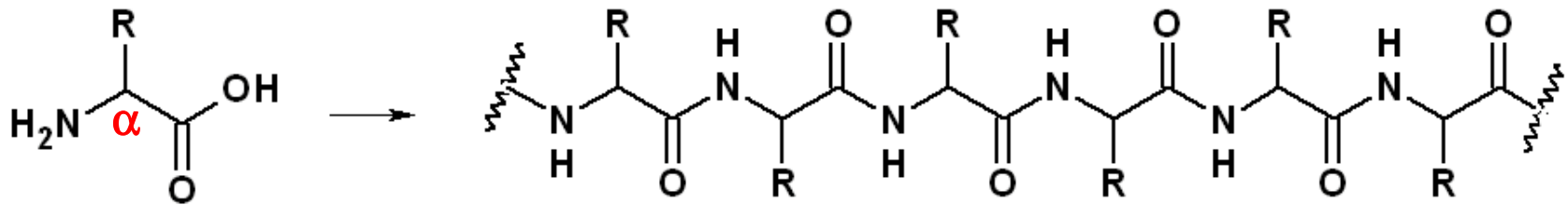
Protein Synthesis: Chemical Ligation

Chemistry and Biology

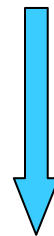


Chemistry and Biology

α -world



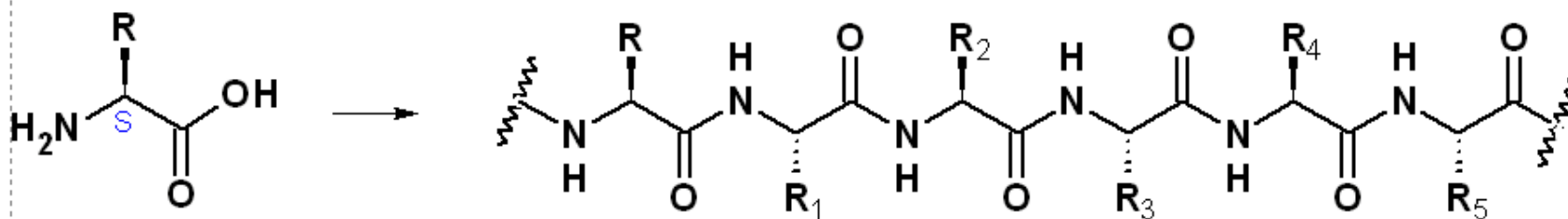
Metabolism



Structure and functions

Chemistry and Biology

α -world

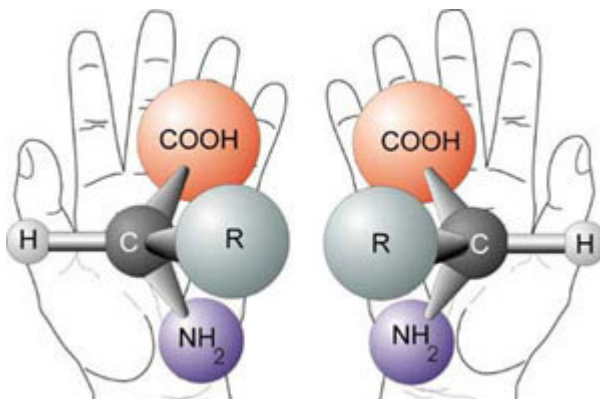


- 20 R have been found in protein
(22 with selenocysteine and pyrrolysine)
- α -Carbon stereochemistry: (S) in protein
Both configuration in many natural products

Chemistry and Biology

α -world

Life and chirality



nature
chemistry

ARTICLES

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Complete chiral symmetry breaking of an amino acid derivative directed by circularly polarized light

Wim L. Noorduin¹, Arno A. C. Bode¹, Maarten van der Meijden², Hugo Meekes¹, Albert F. van Etteger¹, Willem J. P. van Enkevort¹, Peter C. M. Christianen¹, Bernard Kaptein³, Richard M. Kellogg², Theo Rasing¹ and Elias Vlieg^{1*}

Chemistry and Biology

α -world

Life and chirality

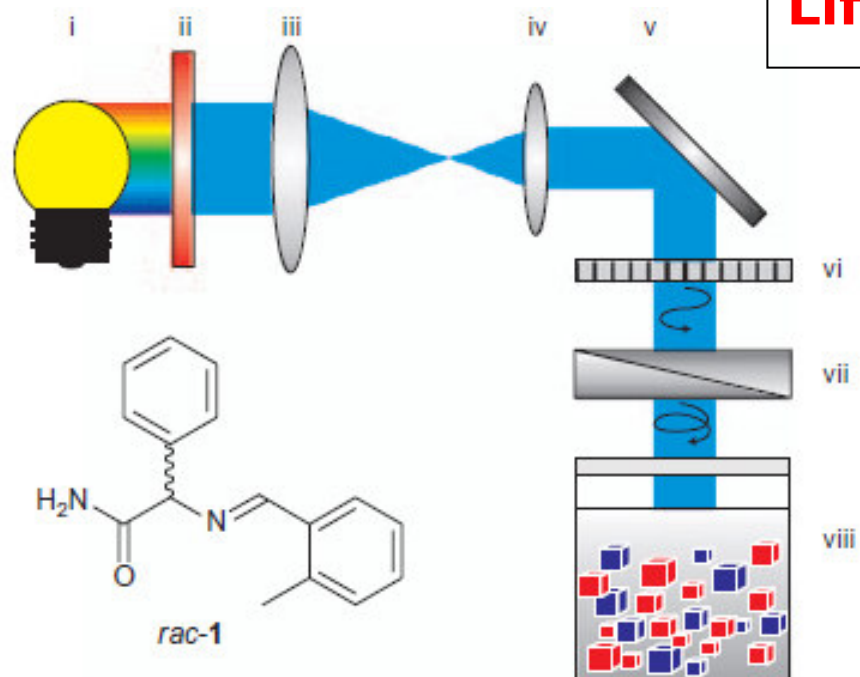


Figure 1 | Experimental set-up for CPL-driven deracemization. Light from a 200 W Hg (Xe) light source (i) is filtered through a 300-400 nm filter (ii), then focused using two quartz lenses (iii, iv). The focused light is then directed using an ultraviolet mirror (v) through a linear polarizer (vi), followed by a Babinet-Soleil compensator (vii) to generate either *r*- or *l*-CPL. The light is used to irradiate solutions and/or suspensions of compound **1** in a reaction vessel (viii) with a quartz window at the top. Inset shows the structural formula of **1**.

Chemistry and Biology

α -world

Life and chirality

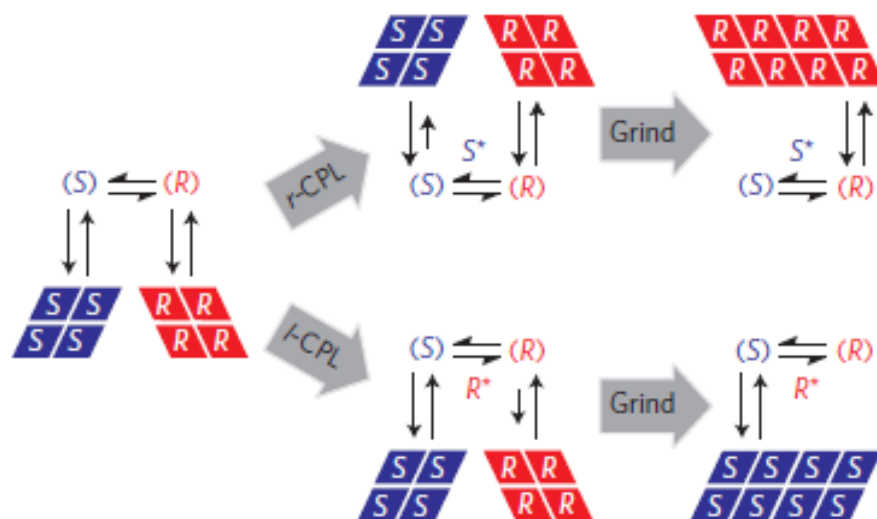
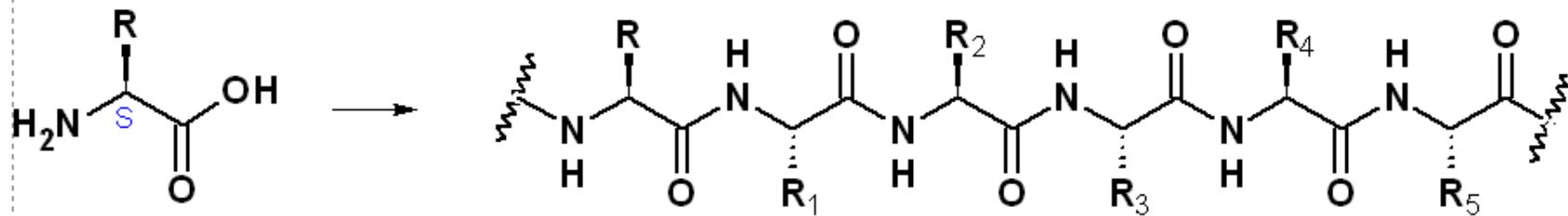


Figure 3 | Cascade of events during the CPL-directed deracemization of racemic 1. Initially, dissolved enantiomers of **1**, represented by (R) and (S), are in equilibrium with the corresponding crystal phase. During the l-CPL irradiation of a solution–solid mixture of racemic **1** an enantioselective growth inhibitor R^* is formed (lower route). This compound R^* enantioselectively hampers the growth of (R)-**1** crystals so that the solid phase enriches in (S)-**1**. This chiral imbalance is fully amplified on abrasive grinding, which results in the complete deracemization of racemic **1** to give a solid phase of (S)-**1**. A similar, but opposite, process occurs on r-CPL irradiation (upper route).

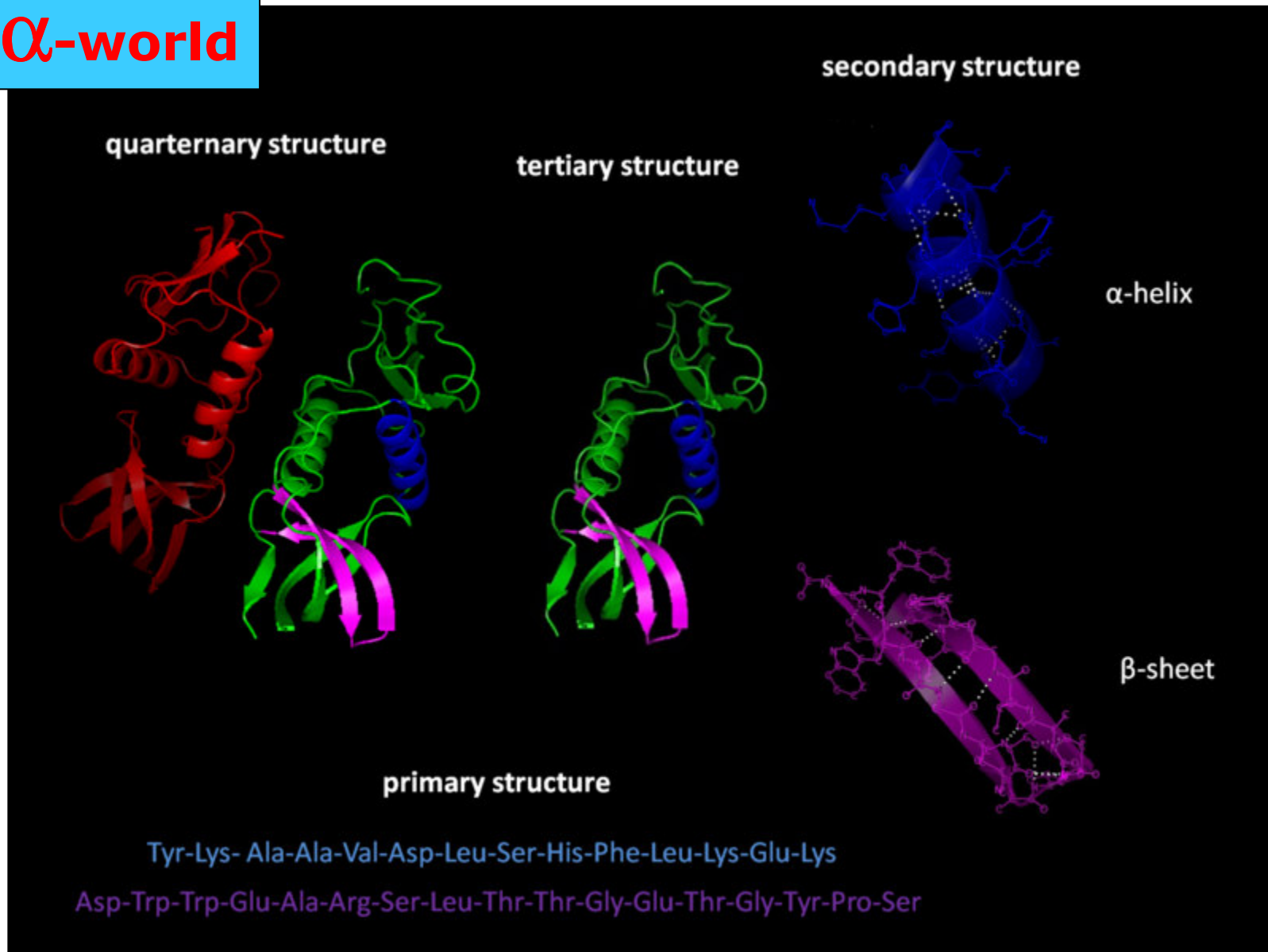
Chemistry and Biology

α -world



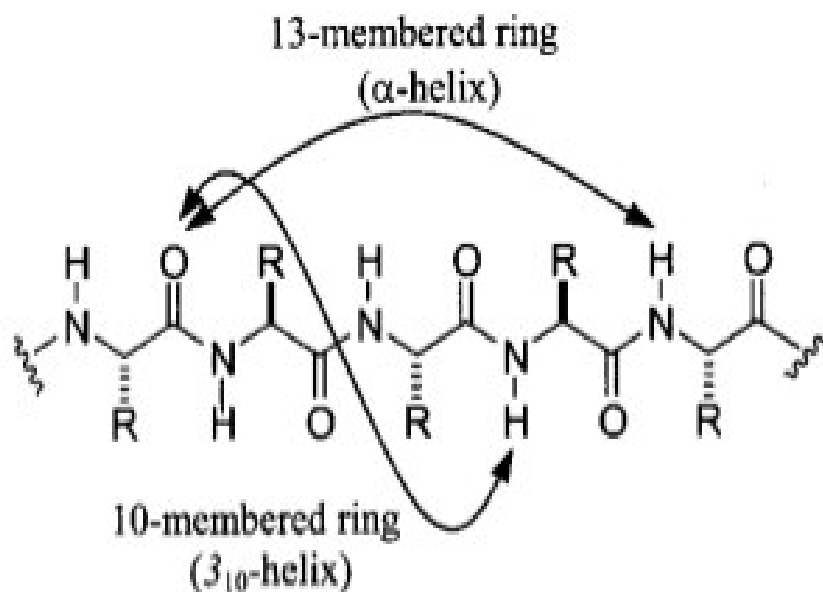
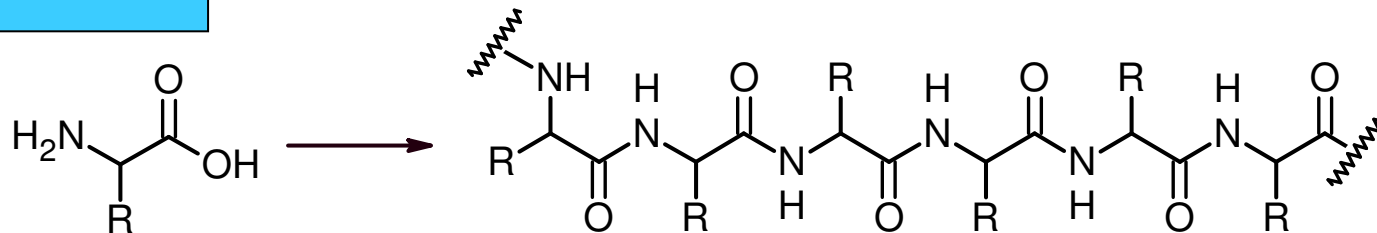
Chemistry and Biology

α -world



Chemistry and Biology

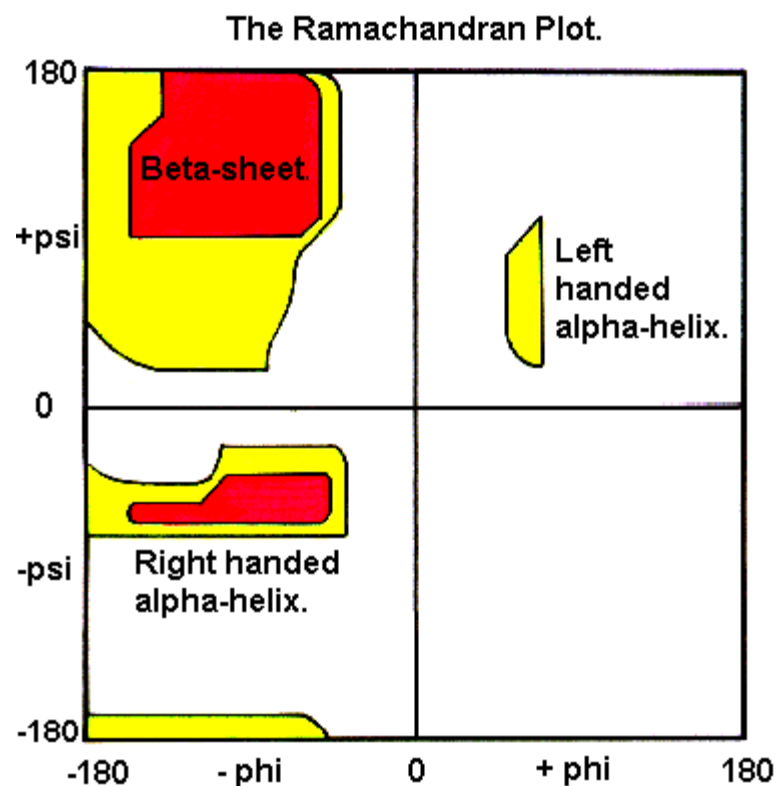
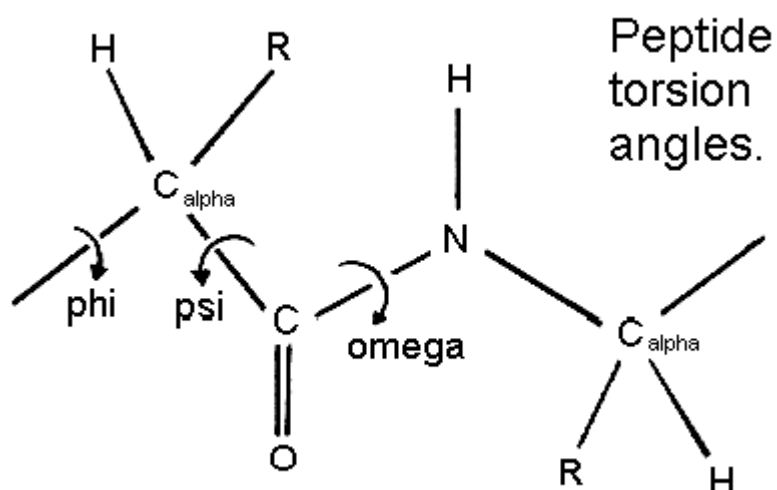
α -world



H-bond patterns associated with helical secondary structures in α -polypeptides

Chemistry and Biology

α -world



Chemistry and Biology

α -world

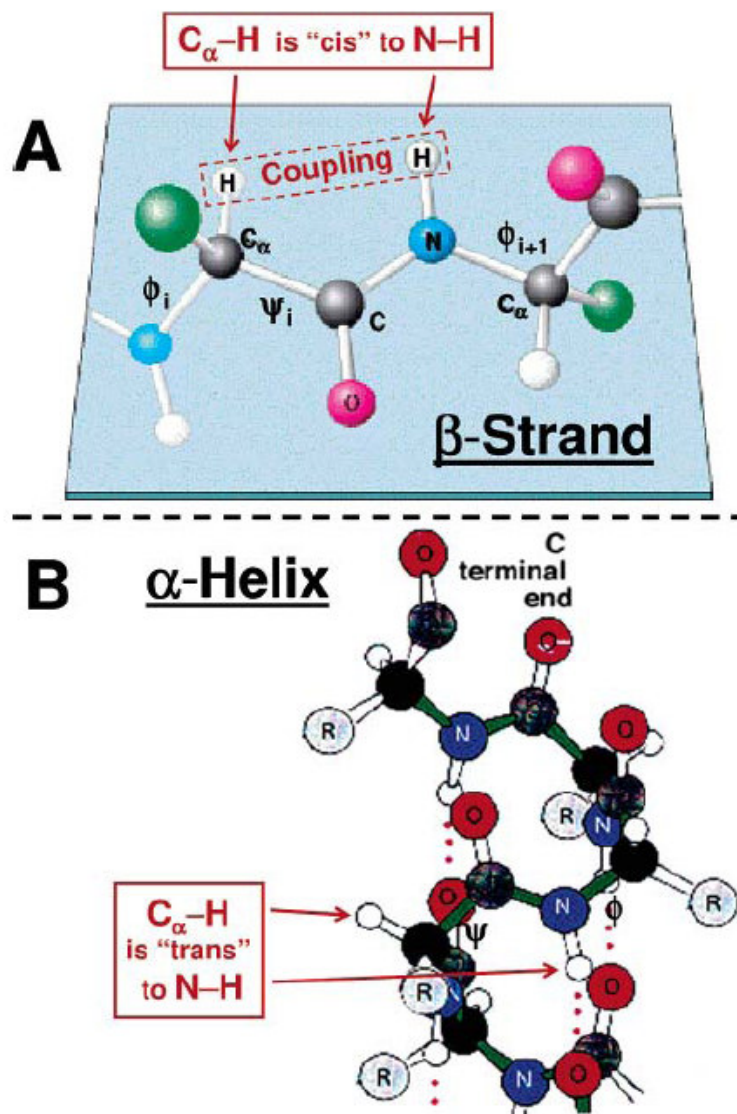
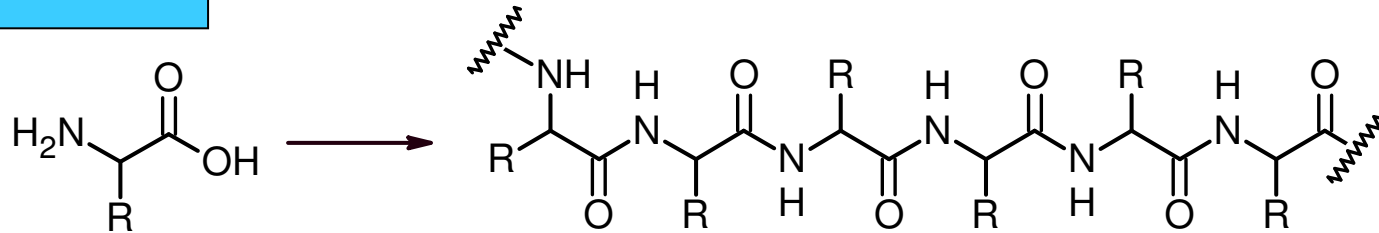


Figure 1. Model of a polypeptide chain (A) at β -strand-like and (B) at α -helix-like Ψ and Φ Ramachandran angles. The distance between $C_\alpha-H$ and $N-H$ hydrogens depends on the Ψ Ramachandran angle.

Chemistry and Biology

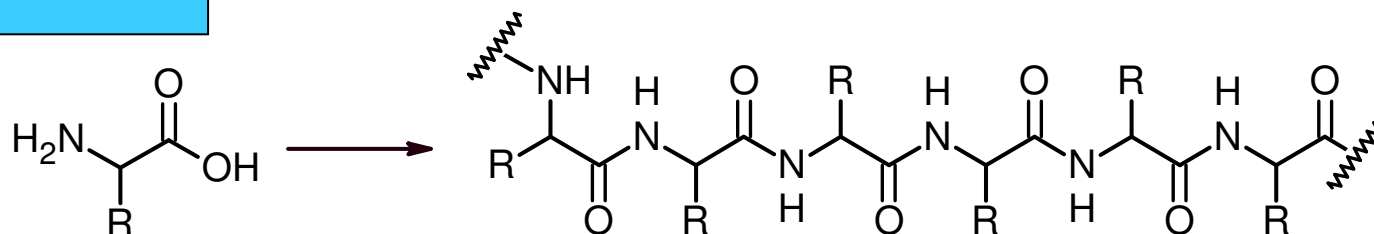
α -world



$$\# \text{protein} = (n)^L \quad \begin{array}{l} n = \# \text{ AA} \\ L = \text{protein length} \end{array}$$

Chemistry and Biology

α -world



$$\# \text{protein} = (n)^L \quad \begin{array}{l} n = \# \text{ AA} \\ L = \text{protein length} \end{array}$$

$$\begin{array}{l} n = 20 \\ L = 100 \end{array} \rightarrow 20^{100} \rightarrow 10^{130} \text{ protein}$$

Complex organism know $< 10^6$?

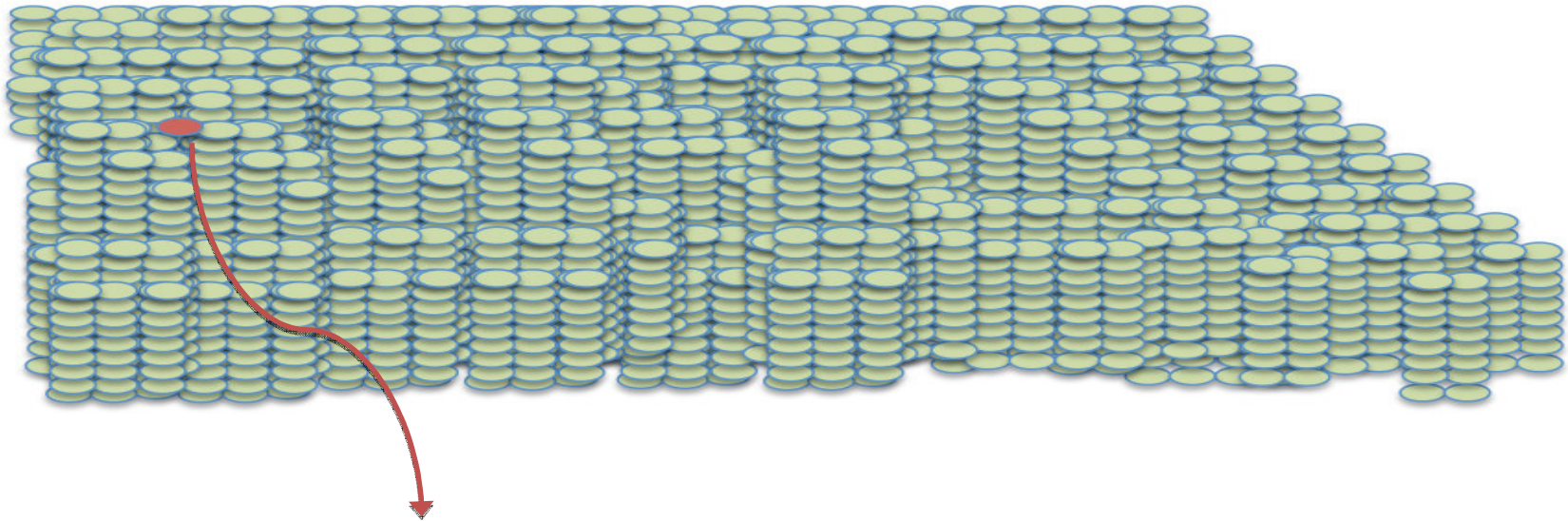
Chemistry and Biology

α -world

$$\begin{aligned} n &= 20 \\ L &= 100 \end{aligned}$$



10^{130} protein



Protein of real world

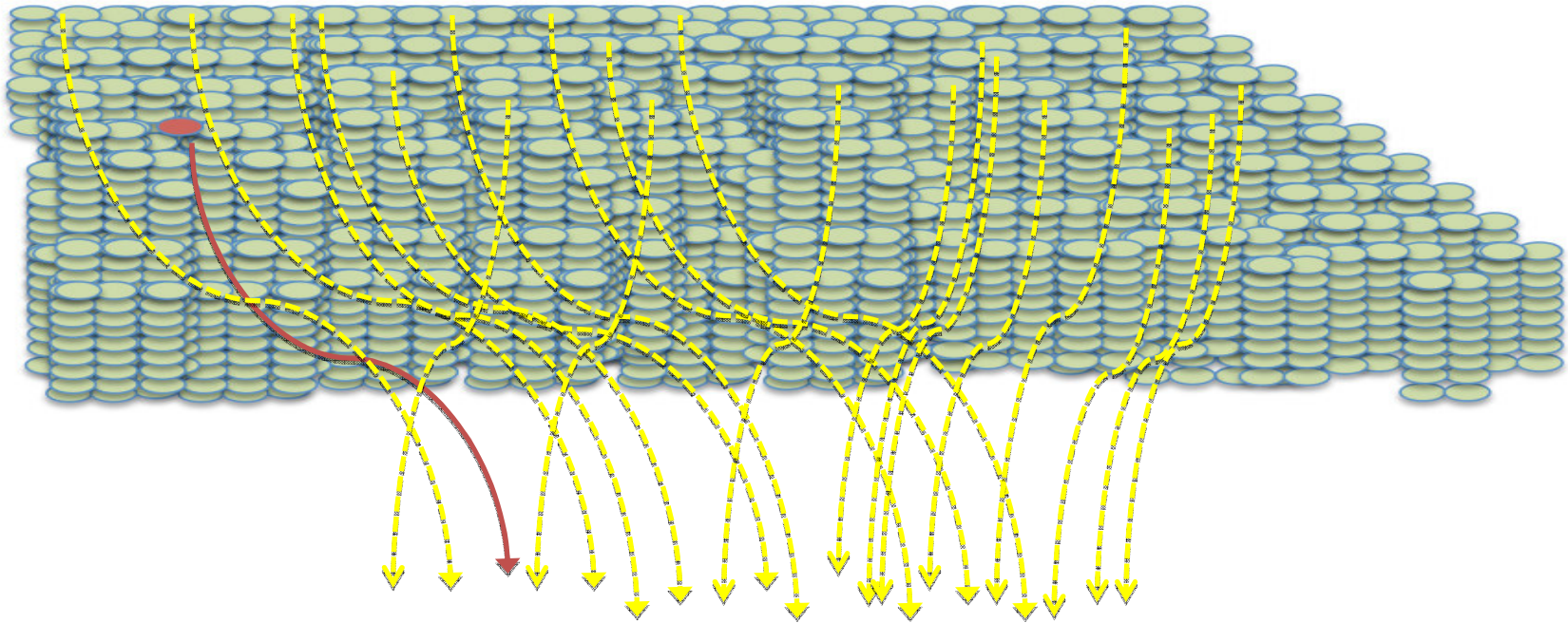
Chemistry

α -world

n = 20
L = 100



10^{130} protein

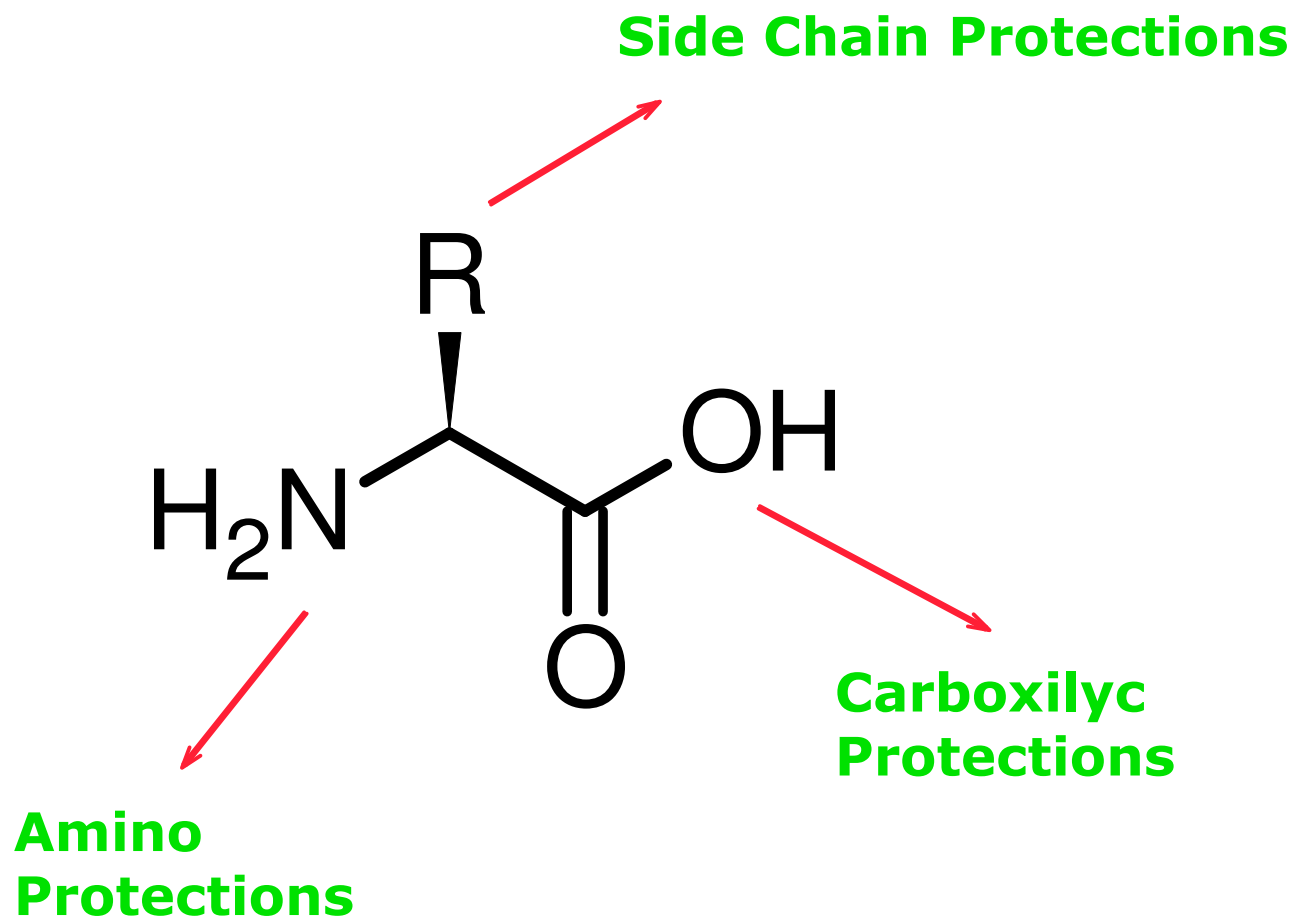


Accessible by chemical synthesis

- SPPS Methods
- Chemical Ligation

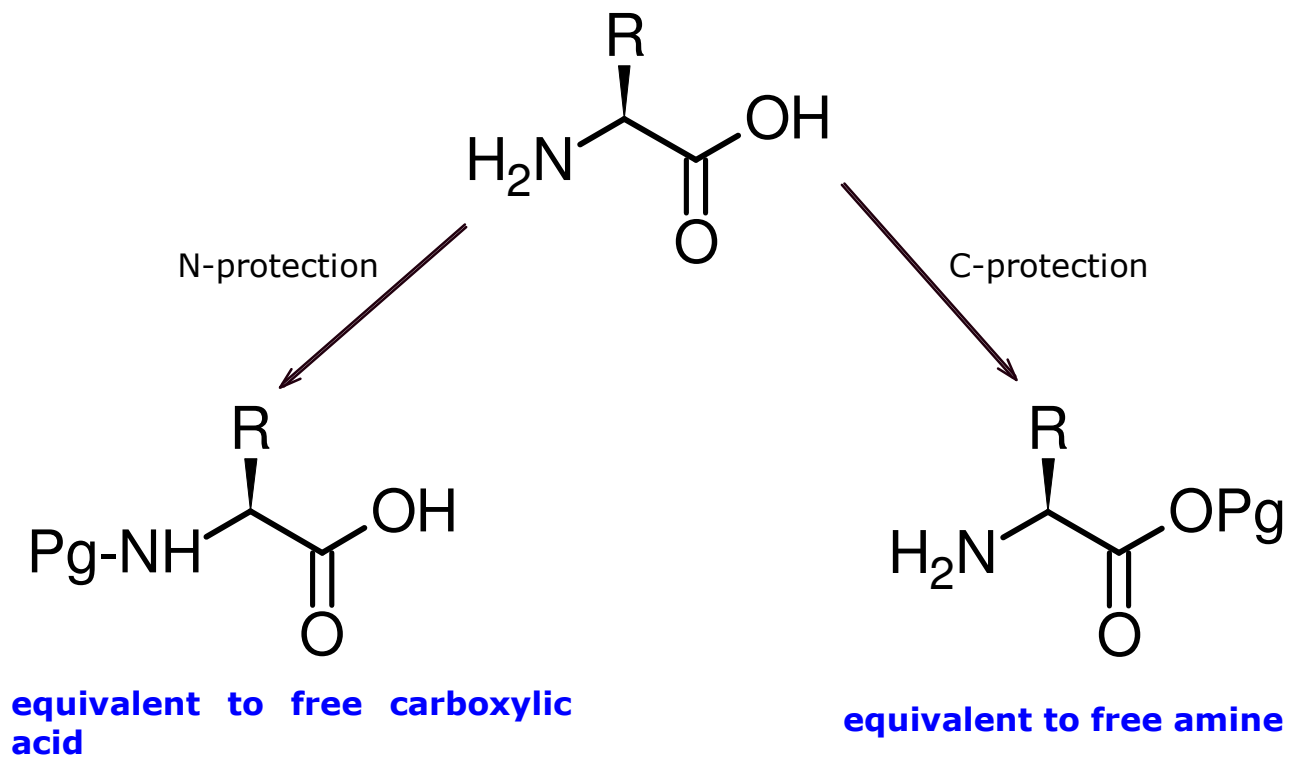
Chemistry

Protecting Groups



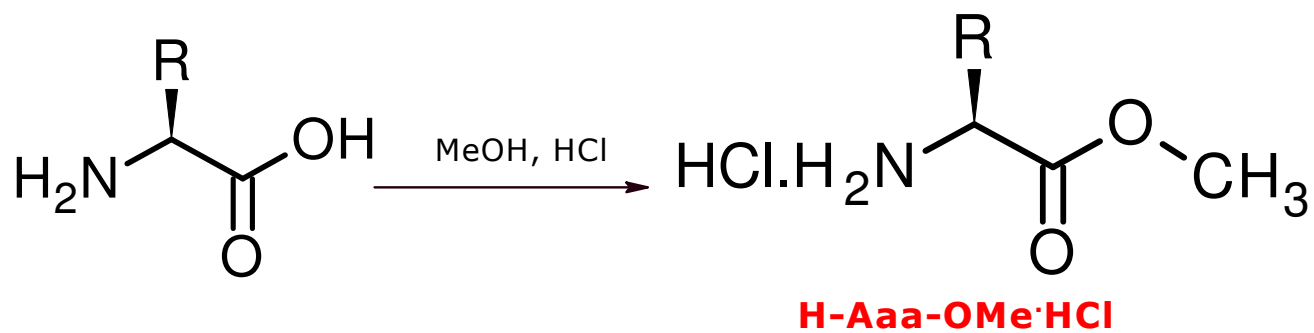
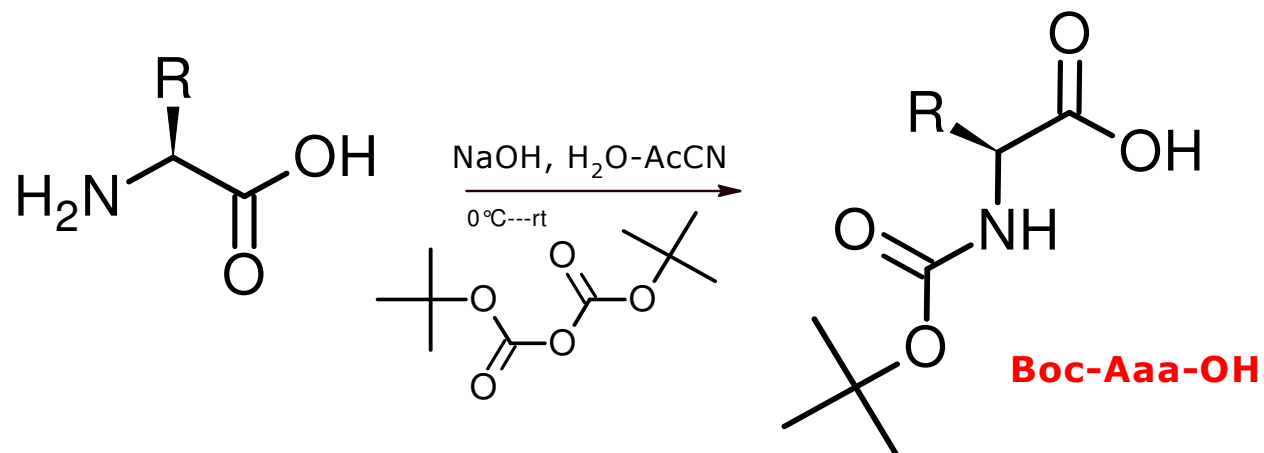
Chemistry

Protecting Groups



Chemistry

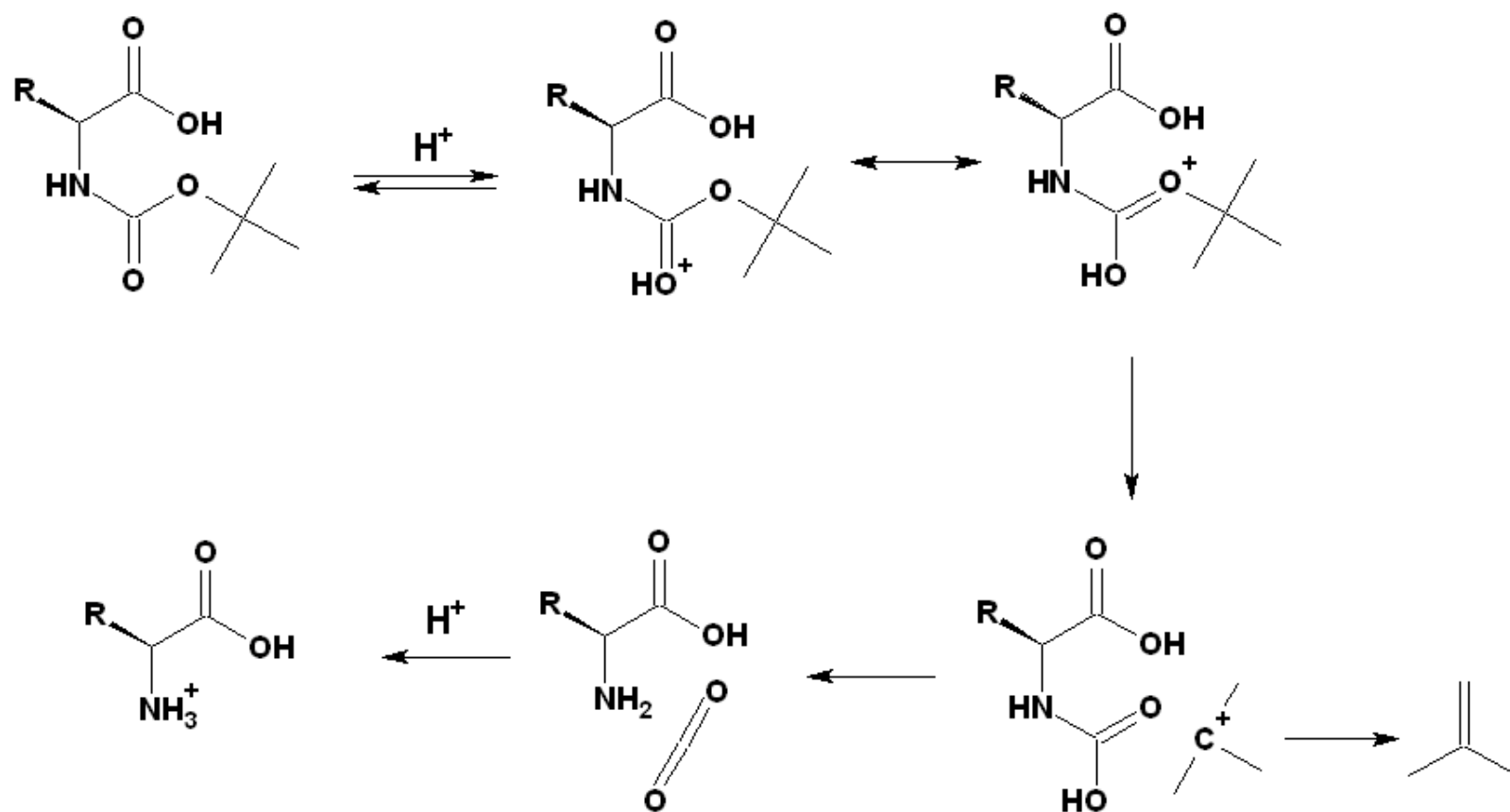
Protecting Groups



Chemistry

Protecting Groups

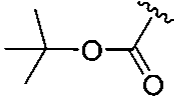
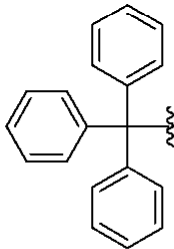
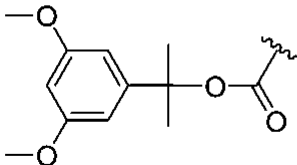
Boc deprotection



Chemistry

Protecting Groups

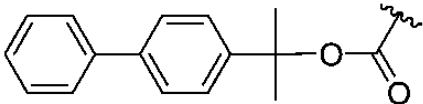
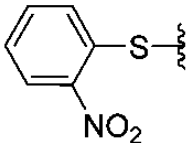
Table. Amino-Protecting Groups Removed by Acid

Name and Structure	Removal conditions	Stability to the removal of
<i>tert</i>-Butyloxycarbonyl (Boc) 	1) 25-50% TFA-DCM 2) 4 M HCl in dioxane 3) 2 M MeSO ₃ H in dioxane 4) 1 M TMS-Cl, 1 M phenol-DCM	Fmoc, Z, ^a Trt, Alloc, <i>p</i> NZ
<i>T</i>rityl (Trt) 	1) 1% TFA-DCM 2) 0.1 M HOBT-TFE 3) 0.2% TFA, 1% H ₂ O-DCM 4) 3% TCA-DCM	Fmoc, Alloc
3,5-Dimethoxyphenylisoproxycarbonyl (Ddz) 	1-5% TFA-DCM	Fmoc, Alloc

Chemistry

Protecting Groups

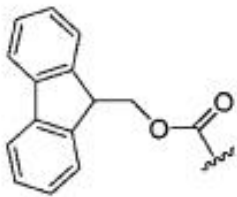
Table. Amino-Protecting Groups Removed by Acid

Name and Structure	Removal conditions	Stability to the removal of
2-(4-Biphenyl)isopropoxycarbonyl (Bpoc) 	0.2-0.5%-TFA	Fmoc, Alloc
2-Nitrophenylsulfenyl (Nps) 	1) Diluted solutions of HCl-CHCl ₃ -AcOH 2) 2-Mercaptopyridine-AcOH-MeOH, DMF or DCM 3) Ni Raney column in DMF	Fmoc

Chemistry

Protecting Groups

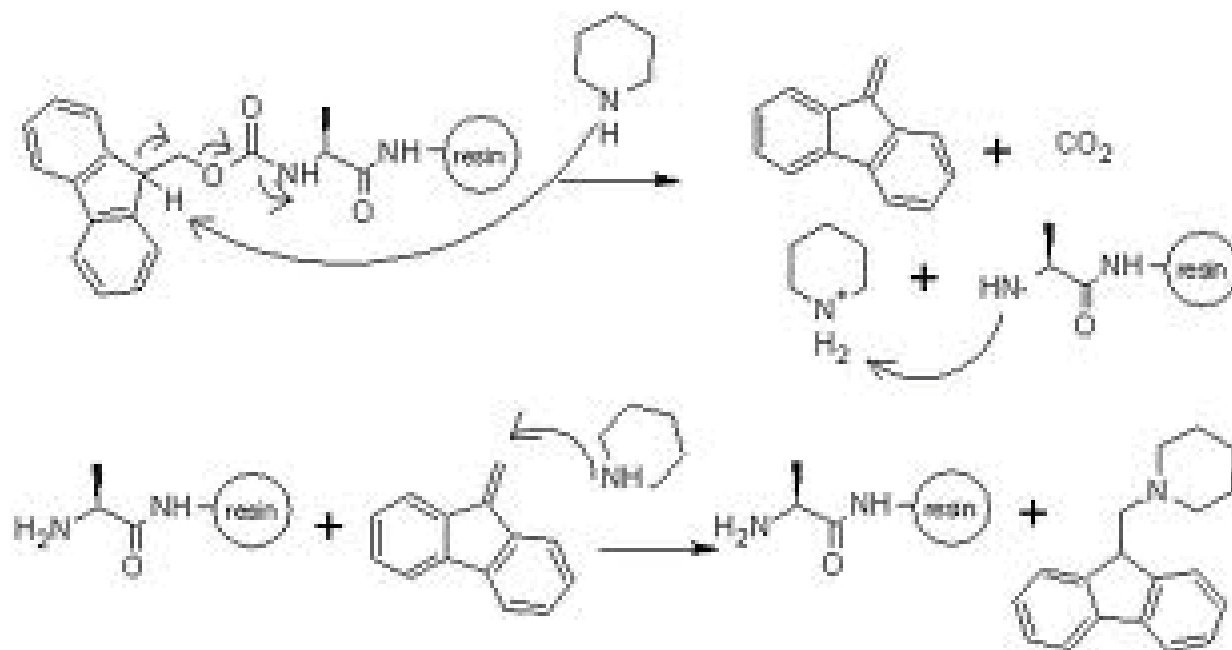
Table. Amino-Protecting Groups Removed by Base

Name and Structure	Removal conditions	Stability to the removal of
9-Fluorenylmethoxycarbonyl (Fmoc) 	<u>Solid phase:</u> 1) 20% piperidine-DMF 2) 1-5% DBU-DMF 3) morpholine-DMF (1:1) 4) 2% HOBt, 2% hexamethyleneimine, 25% <i>N</i>-methylpyrrolidine in DMSO-NMP (1:1) <u>Solution:</u> 1) NH₃ (10 h) 2) morpholine or piperidine in organic solvents (within minutes) 3) 10% DEA, DMA (2 h) 4) polymeric secondary amines (i.e. piperidine, piperazines) in organic solvents	Boc, Z,^a Trt, Alloc, <i>p</i>NZ^a

Chemistry

Protecting Groups

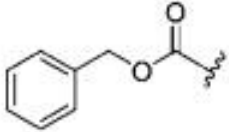
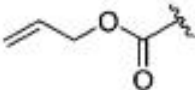
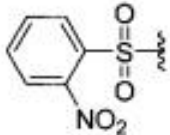
Fmoc
deprotection



Chemistry

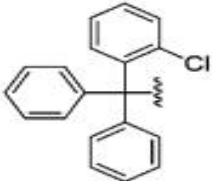
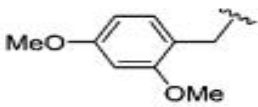
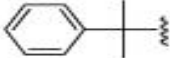
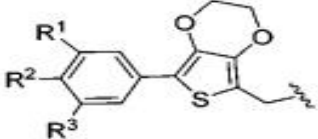
Protecting Groups

Table. Other Amino-Protecting Groups

Name and Structure	Removal conditions	Stability to the removal of
<i>Benzyloxycarbonyl (Z)</i> 	1) H ₂ cat 2) Strong acids such as: HBr in AcOH, TFA at high temperatures, TFA-thioanisole or liquid HF 3) BBr ₃	Boc, Fmoc, Trt, Alloc, <i>p</i> NZ ^a
<i>Allyloxycarbonyl (Alloc)</i> 	Pd(PPh) ₃ cat., scavengers: H ₃ N·BH ₃ , Me ₂ NH·BH ₃ or PhSiH ₃ in organic solvents	Boc, Fmoc, Trt, <i>p</i> NZ ^a
<i>o</i>-Nitrobenzenesulfonyl (<i>o</i>NBS) 	1) 5% thiophenol-DMF, 2 eq. of K ₂ CO ₃ (primary amines) 2) β-mercaptoethanol and DBU (secondary amines)	Boc, Fmoc, Trt

Chemistry

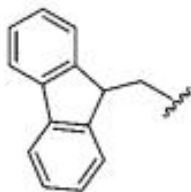
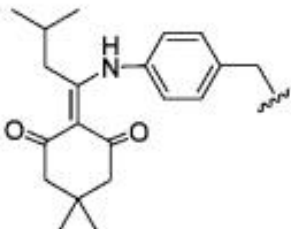
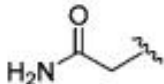
Protecting Groups Table. Carboxylic Acid-Protecting Groups Removed by Acid

Name and Structure	Removal conditions	Stability to the removal of
<i>tert</i>-Butyl (<i>t</i>Bu) 	90% TFA-DCM (solid phase and solution) or 4 M HCl in dioxane (solution)	Fmoc, Z, ^a Trt Alloc, <i>p</i> NZ,
2-Chlorotrityl (2-Cl-Trt) 	1% TFA-DCM	Fmoc, Alloc
2,4-Dimethoxybenzyl (Dmb) 	1% TFA-DCM	Fmoc, Alloc
2-Phenylisopropyl (2-PhⁱPr) 	4% TFA-DCM	Fmoc, Alloc
5-Phenyl-3,4-ethylenedioxythienyl (Phenyl-EDOTn)  R¹=R²=R³= OMe; R¹=R²= OMe, R³= H; R¹=R²= H, R³= OMe or R¹=R²=R³= H.	0.01%-0.5% TFA-DCM and scavengers	Fmoc

Lgl12LD1

Chemistry

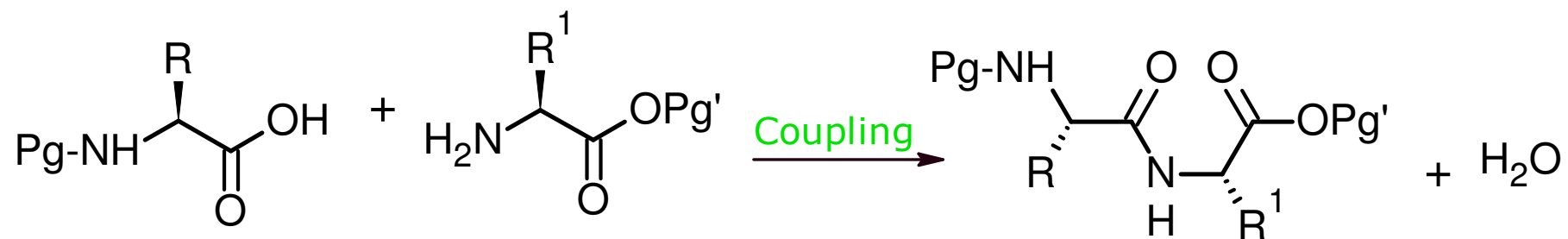
Protecting Groups Table. Carboxylic Acid-Protecting Groups Removed by Base

Name and Structure	Removal conditions	Stability to the removal of
9-Fluorenylmethyl (Fm) 	15% DEA or 20% piperidine-DMF or DCM	Boc, Trt, Alloc
4-(N-[1-(4,4-dimethyl-2,6-dioxocyclohexylidene)-3-methylbutyl]-amino)benzyl (Dmab) 	2% hydrazine-H ₂ O-DMF (1:1)	Boc, Fmoc, Trt,
Methyl (Me) and Ethyl (Et)	LiOH, NaOH or KOH	Boc, Z
Carbamoylmethyl (Cam) 	NaOH or Na ₂ CO ₃ -DMF-H ₂ O	Boc, Fmoc ^a Z ^b

Chemistry

Peptide Coupling

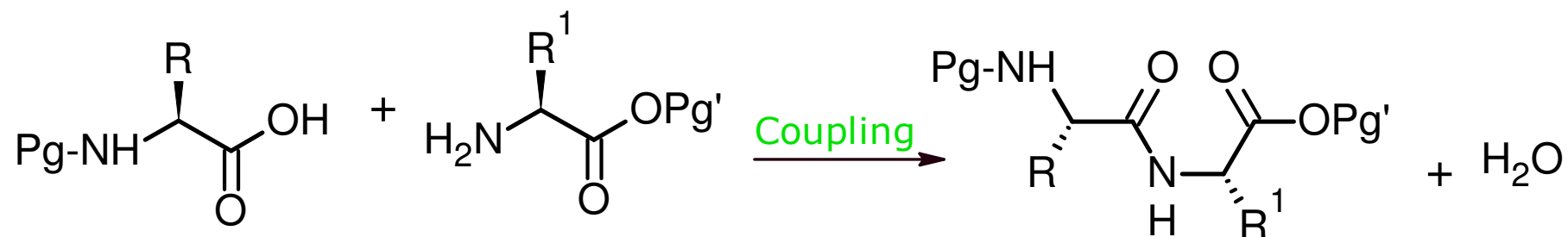
Solution synthesis



Chemistry

Peptide Coupling

Solution synthesis



- Mixed anhydrides
- Carbodiimides
- Phosphonium salts
- Uronium salts

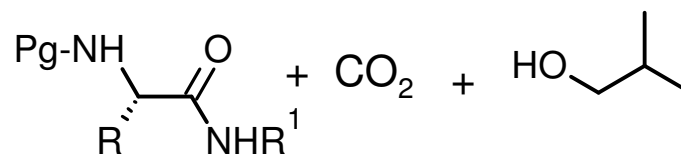
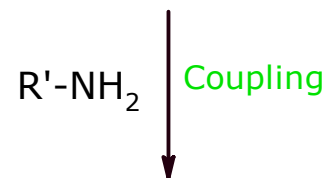
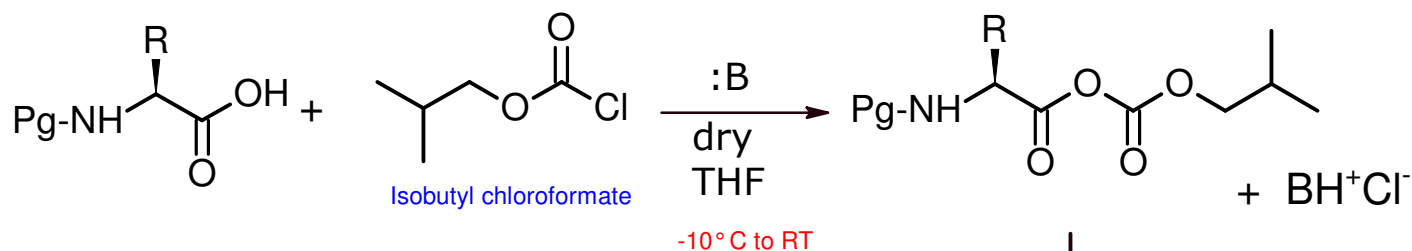
Chemistry

Peptide Coupling

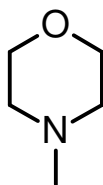
Solution synthesis

•Mixed anhydrides

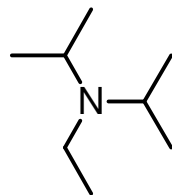
- Carbodiimide
- Phosphonium salt
- Uronium salt



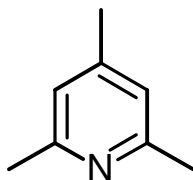
:B =



NMM



DIPEA

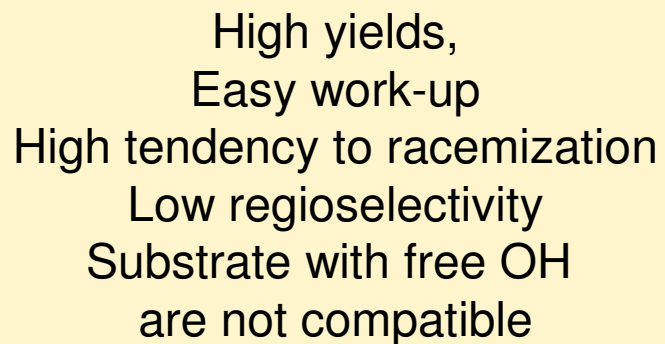


sym-collidine

Fast process
High yields,
Easy work-up
Virtually racemization free
High regioselectivity
Substrate with free OH
are compatible

Peptide Coupling

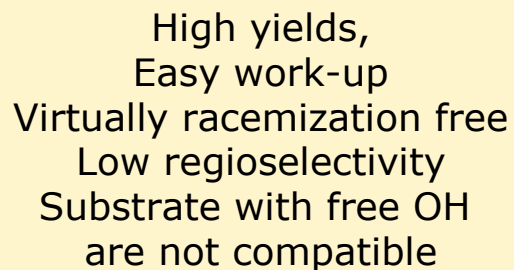
- Mixed anhydrides
- Carbodiimides**
- Phosphonium salts
- Uronium salts



Peptide Coupling

Active esters

- Mixed anhydrides
- **Carbodiimides**
- Phosphonium salts
- Uronium salts

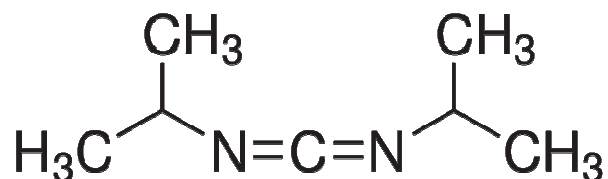


Chemistry

Peptide Coupling

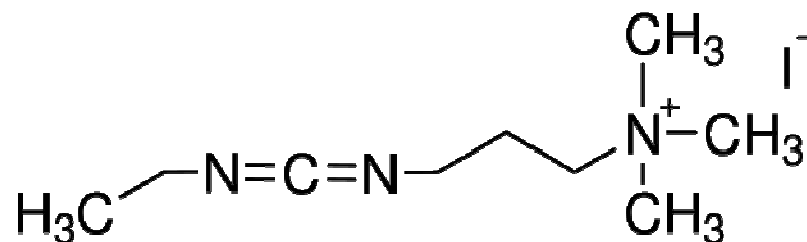
Solution synthesis

- Mixed anhydrides
- Carbodiimides**
- Phosponium salts
- Uronium salts



***N,N'* -Diisopropylcarbodiimide**

Used specially in SPPS for attachment of first residue



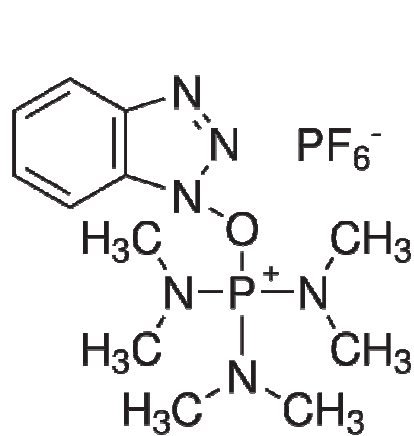
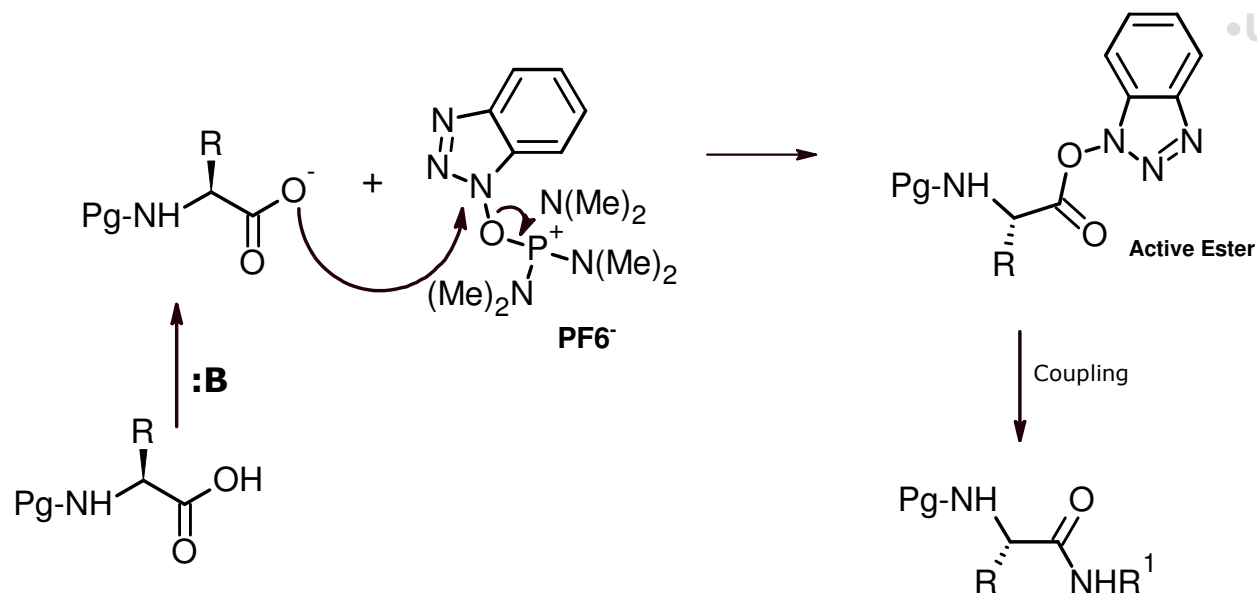
1-[3-(Dimethylamino)propyl]-
3-ethylcarbodiimide methiodide
Water-soluble carbodiimide

Chemistry

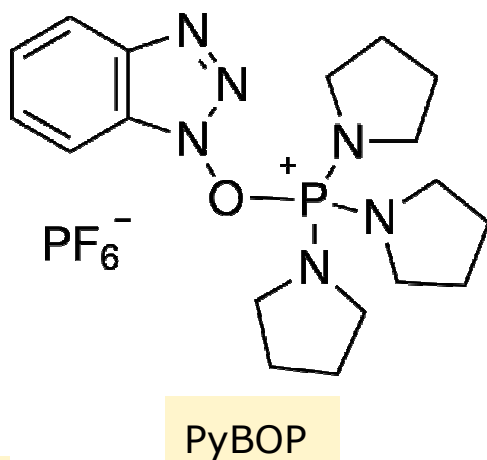
Peptide Coupling

Solution $\square$ **solid-phase synthesis**

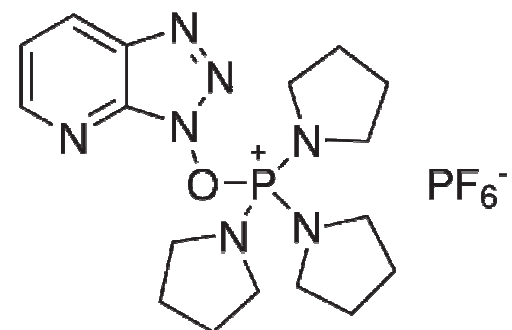
- Mixed anhydrides
- Carbodiimides
- **Phosphonium salts**
- Uronium salts



BOP Reagent, BOP, Castros reagent



PyBOP



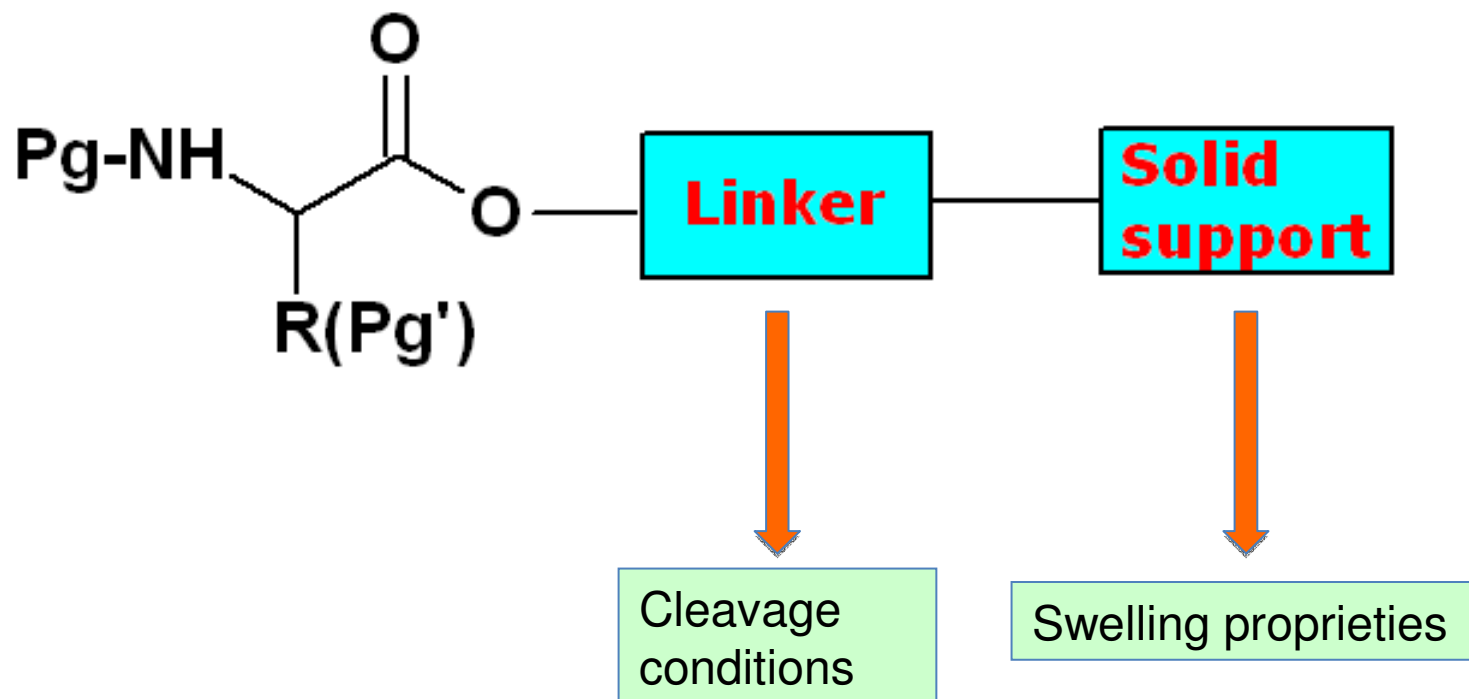
PyAOP

Chemistry

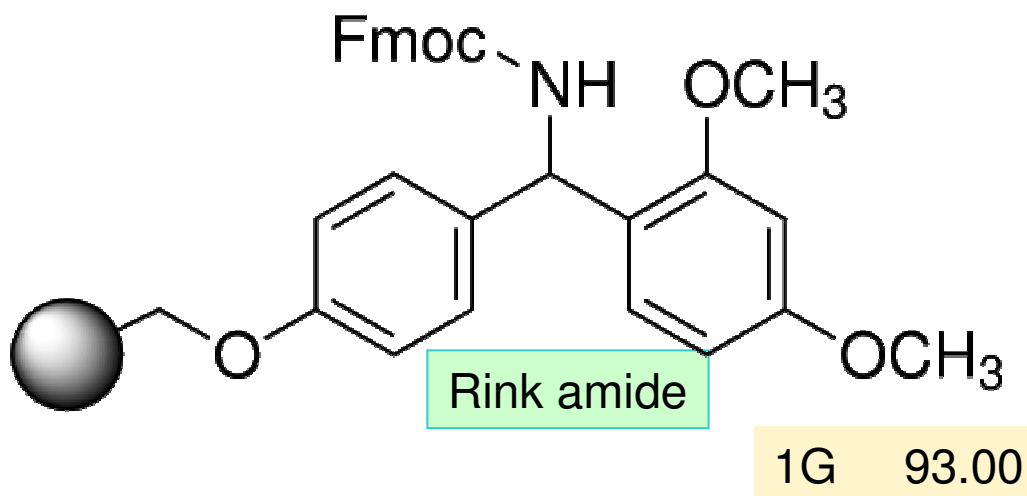
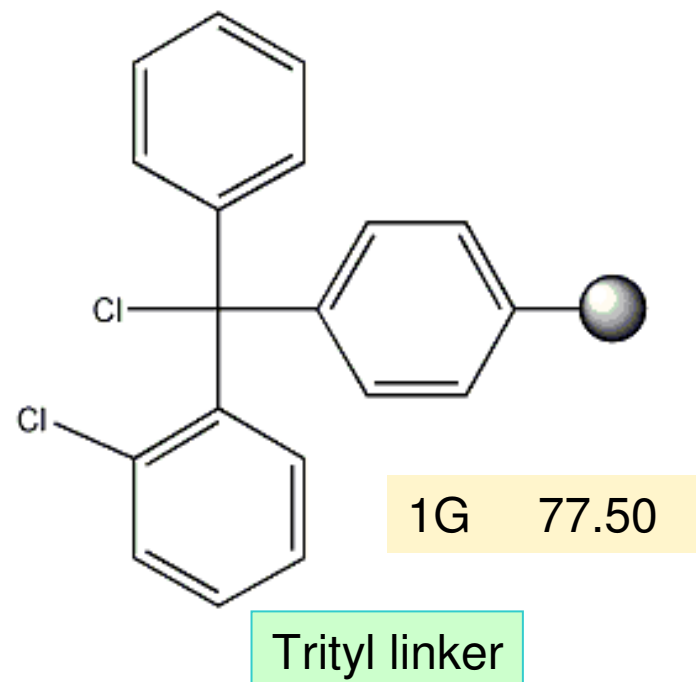
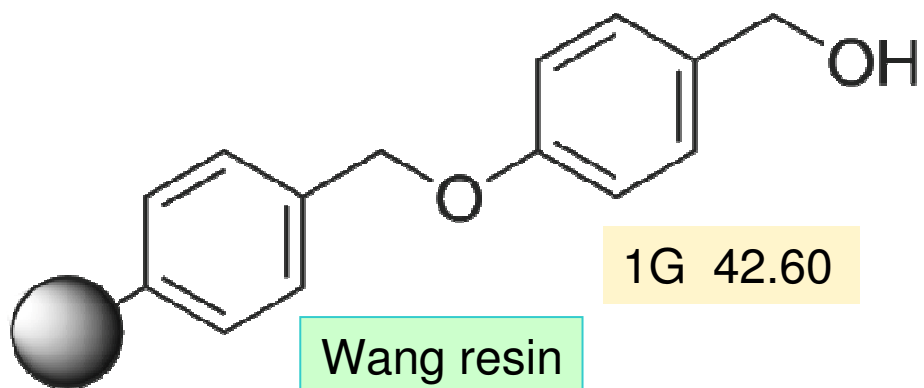
Peptide Coupling

solid-phase synthesis

- Mixed anhydrides
- Carbodiimide
- Phosphonium salts
- Uronium salt



Chemistry

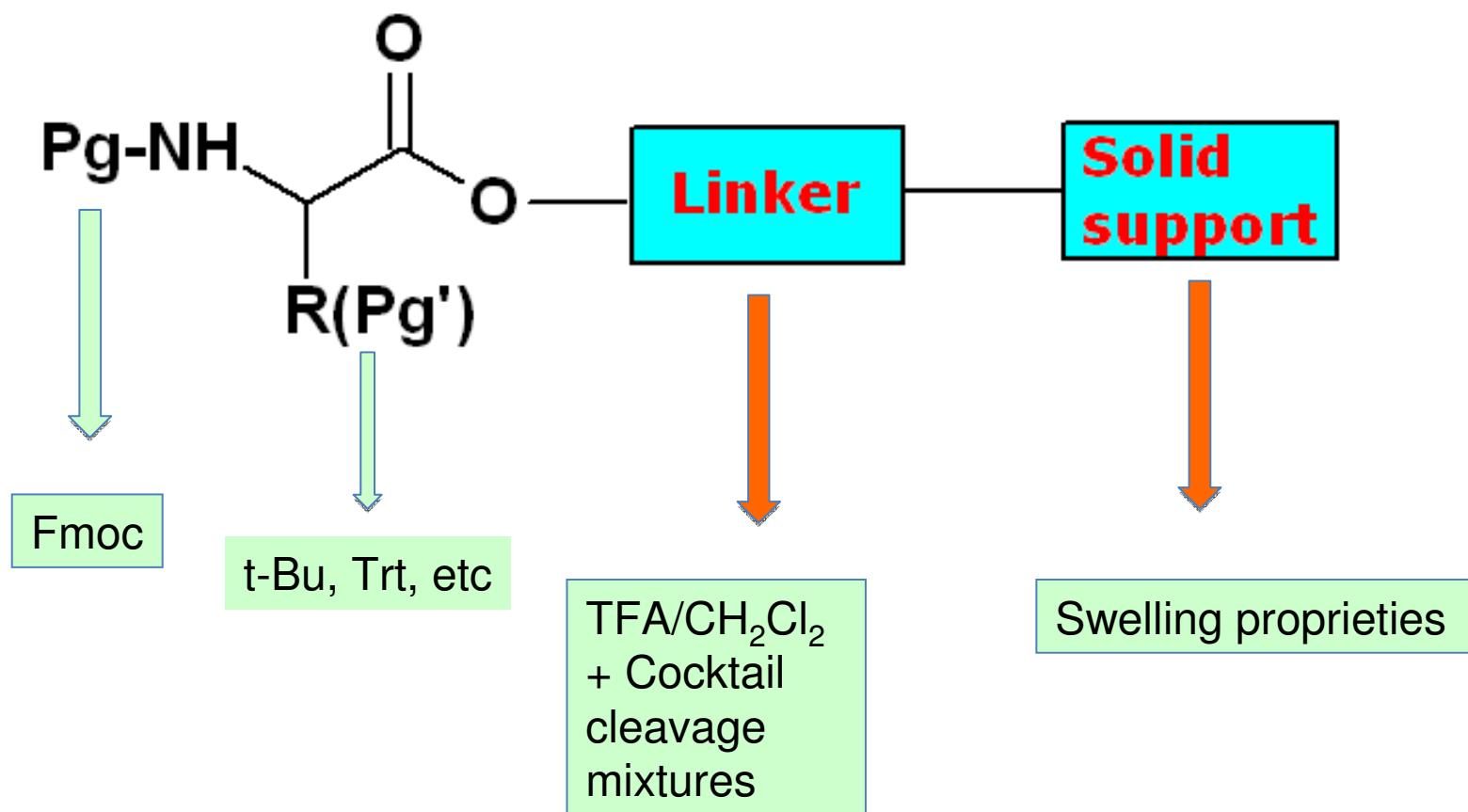


Chemistry

Peptide Coupling

solid-phase synthesis

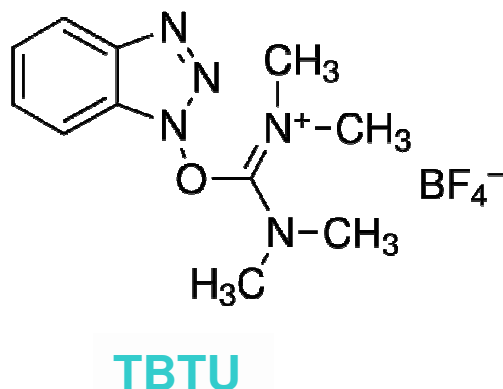
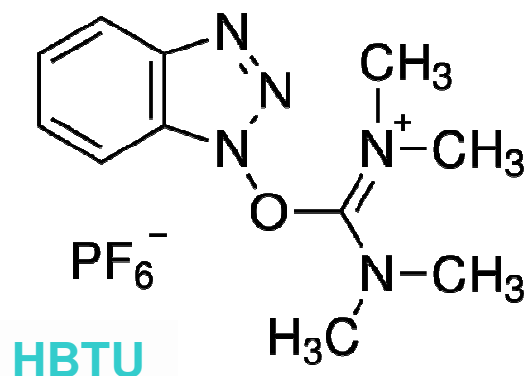
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- Carbodiimides
- Phosphonium salts
- Uronium salts



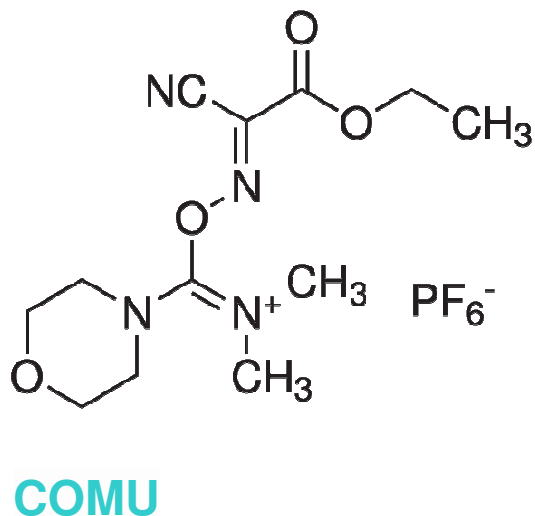
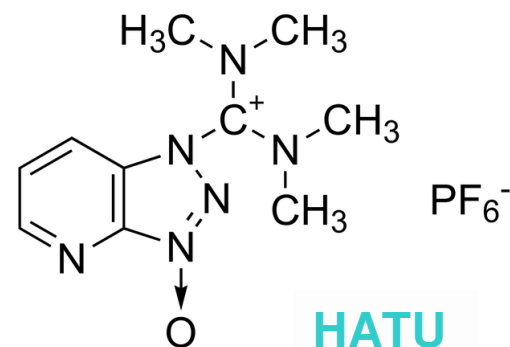
Chemistry

Peptide Coupling

solid-phase synthesis



- Mixed anhydrides
- Carbodiimides
- Phosphonium salts
- **Uronium salts**



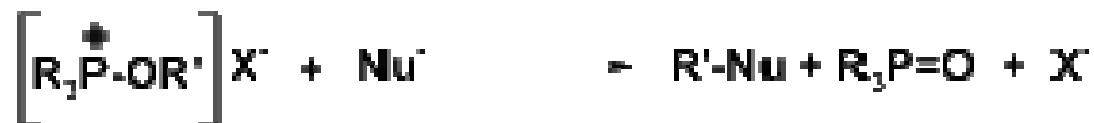
Advantages

- Equal or even superior performance to HATU
- Non-explosive (does not contain benzotriazole moiety)
- Suitable for solution phase & solid phase peptide synthesis
- Retention of configuration – low to non-existent racemization
- High solubility and stability in typical solvents
- Visual or colorimetric reaction monitoring possible
- Easy removal of water-soluble by-products

Chemistry

Peptide Coupling

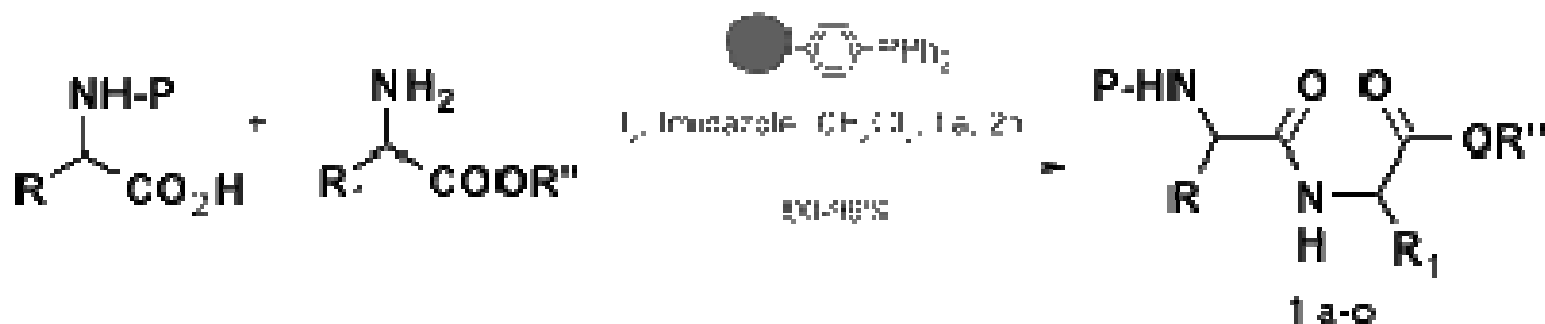
- Mixed anhydrides
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Chemistry

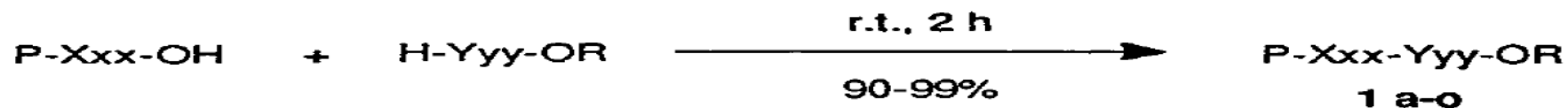
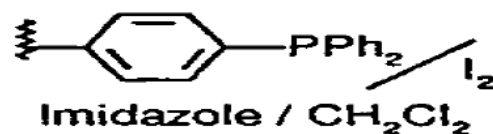
Peptide Coupling

- Mixed anhydrides
- Carbodiimides
- Phosponium salts
- Uronium salts



Chemistry

Peptide Coupling



1	P	Xxx	Yyy	R
a	Fmoc	L-Ala	L-Leu	All
b	Cbz	L-Met	L-Ala	Me
c	Fmoc	L-Cys(Trt)	L-Cys(Trt)	Me
d	Boc	L-Leu	L-Phe	Bu ^f
e	Fmoc	L-Phe	Gly	Bu ^f
f	Fmoc	D,L-Phe	Gly	Bu ^f
g	Fmoc	L-Val	L-Val	All
h	Fmoc	L-Phe	L-Leu	Bn
i	Boc	L-Trp	L-Leu	Me
l	Fmoc	L-Glu(Bu ^f)	L-Phe	All
m	Cbz	L-Ala	L-Phe	All
n	Boc	L-Arg(Ts)	L-Phe	Me
o	Cbz	L-Thr(OH)	L-Val	Hept

Chemistry

Peptide Coupling

- Mixed anhydrides
- Carbodiimide
- Phosphonium salt
- Uronium salts

Solid-phase synthesis steps:

1. Attachment of first residue Fmoc-Aaa-OH, DIC, DMF
2. Fmoc deprotection 20% Piperidine in DMF
3. Coupling Fmoc-Aaa-OH, DIPEA, HBTU, HOBt, DMF
4. Iterative steps 2 and 3
5. Cleavage TFA in CH_2Cl_2 , Thianisole, Triethylsilane,
Precipitation in dry ether
6. Analysis and purification HPLC/MS, Preparative HPLC

Chemistry

Peptide Coupling

Primary Sequence of Recombinant α -gliadine

V¹RVPVPQLQPQNPSQQQPQE²⁰QVPLVQQQQQF
LGQQQQPFPPQ⁴⁰QPYPQPQPFPSQQQPQLQLQP⁶⁰
FPQPQLPYSQPQPFRPQQPY⁸⁰PQPQPQYSQPQ
QPISQQQQQQ¹⁰⁰QQQQQQQQQQQQQQQQQQILQQI
¹²⁰LQQQLIPCMDVVLQQHNIAH¹⁴⁰GRSQVLQQ
STYQLLQELCCQ¹⁶⁰HLWQIPEQSQCQAIHKVVH
A¹⁸⁰IILHQQQKQQQQQPSSQVSFQ²⁰⁰QPLQQYP
LGQGSFRPSQQNP²²⁰QAQGSVQPQQLPQFEEI
RNL²⁴⁰ALQTLPAMCNVYIPPYCTIT²⁶⁰PFGIFGT
N²⁶⁸

Chemistry

Peptide Coupling

Primary Sequence of Recombinant α -gliadine

VRVPVPQLQPQNPSQQQPQE QVPLVQQQQQFLG
QQQPFPQQPYPPQPPFPSQQPQLQLQPPFPQPQ
LPYSQPQPFRPQQPYPPQPPQYSQPQQPISQQQ
QQQQQQQQQQQQQQQQQQQQILQQILQQQLIPCM
DVVLQQHNIAHGRSQVLQQSTYQLLQELCCQH
LWQIPEQSQCQAIHKVVHAILHQQQKQQQQQP
SSQVSFQQPLQQYPLGQGSFRPSQQNPQAQGS
VQPQQLPQFEEIRNLALQTLPAMCNVYIPPYCTIT
PFGIFGTN

Chemistry

Peptide Coupling

Primary Sequence of Recombinant α -gliadine

VRVPVPQLQPQNPSQQQPQE**QVPLVQQQQFLG**
QQQPFPQQPYPQPQPFPSQQPQLQLQPFPQPQ
LPYSQPQPFRPQQPYPQPQPQYSQPQQPISQQQ
QQQQQQQQQQQQQQQQQQQILQQILQQQLIPCM
DVVLQQHNIAHGRSQVLQQSTYQLLQELCCQH
LWQIPEQSQCQAIHKVVHAILHQQQKQQQQQP
SSQVSFQQPLQQYPLGQGSFRPSQQNPQAQGS
VQPQQQLPQFEEIRNLALQTLPAMCNVYIPPYCTIT
PFGIFGTN

Chemistry

Peptide Coupling

Primary Sequence of Recombinant α -gliadine

VRVPVPQLQPQNPSQQQPQEQVPLVQQQQQFLG
QQQPFPFPQQPYPQPQPFPSQQPQLQLQPFPQPQ
LPYSQPQPFRPQQPYPQPQPQYSQPQQPISQQQ
QQQQQQQQQQQQQQQQQQQQILQQILQQQLIPCM
DVVLQQHNIAHGRSQVLQQSTYQLLQELCCQH
LWQIPEQSQCQAIHKVVHAILHQQQKQQQQQP
SSQVSFQQPLQQYPLGQGSFRPSQQNPQAQGS
VQPQQQLPQFEEIRNLALQTLPAMCNVYIPPYCTIT
PFGIFGTN

Chemistry

Peptide Coupling

Solid-Phase Peptide Synthesis
SPPS

26 PEPTIDES
L= 20

**1 operator + 2
2 months
< 8000 €**

Tabella 5: Dati analitici dei peptidi sintetizzati

Peptide		Sequenza	Peso molecolare calcolato	Peso molecolare osservato
P1	1-20	VRVPVPQLQPQNPSQQQPQE	2296.192	2297.28
P2	10-29	PQNPSQQQPQEQVPLVQQQQ	2328.146	2329.18
P3	20-39	EQVPLVQQQQFLGQQQPFPF	2335.196	2336.23
P4	30-49	FLGQQQPFPQPYPQPQPF	2368.164	2369.22
P5	40-59	QQPYPQPQPFPSQQPYLQLQ	2411.191	2412.25
P6	50-69	PSQQPYLQLQFPQPQLPYS	2355.190	2356.30
P7	60-79	PFPQPQLPYSQPQPFRPQQP	2376.201	2377.25
P8	70-89	QPQPFRPQQPYPQPQPQYSQ	2438.170	2439.26
P9	80-99	YPQPQPQYSQPQQPISQQQQ	2397.135	2398.18
P10	90-109	PQQPISQQQQQQQQQQQQQQ	2461.169	2462.18
P11	100-119	QQQQQQQQQQQQQQQQQILQQ	2549.233	2550.15
P12	111-129	QQQQQQQILQQILQQQLIPCM	2436.260	2437.26
P13	120-139	ILQQQLIPCMDVVLQQHNI	2303.210	2304.30
P14	130-149	DVVLQQHNIHGRSQVLQQS	2256.172	2257.28
P15	141-160	GRSQVLQQSTYQLLQELCCQ	2324.125	2325.17
P16	150-169	TYQLLQELCCQHLLWQIPEQS	2459.161	2460.16
P17	160-179	QHLWQIPEQSQCQAIHKVVH	2408.217	2409.20
P18	170-189	QCQAIHKVVHAILHQQQKQ	2349.280	2350.31
P19	180-199	AAILHQQQKQQQQQPSSQVSF	2322.208	2323.26
P20	190-209	QQQPSSQVSFQQPLQQYPLG	2287.123	2288.15
P21	200-219	QQPLQQYPLGQGSFRPSQQN	2300.130	2301.24
P22	210-229	QGSFRPSQQNPQAQGSVQPQ	2168.030	2169.25
P23	220-239	PQAQGSVQPQQLPQFEEIRN	2293.140	2294.34
P24	230-249	QLPQFEEIRNLALQTLPAMC	2314.181	2315.32
P25	240-259	LALQTLPAMCNVYIPPYCTI	2223.114	2224.15
P26	250-268	NVYIPPYCTITPFGIFGTN	2116.034	2117.06