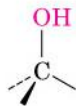
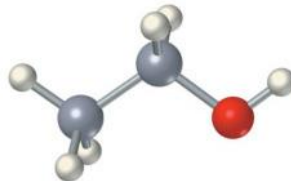
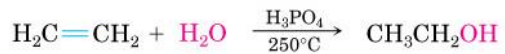
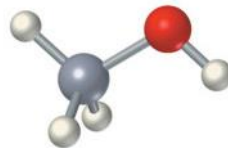
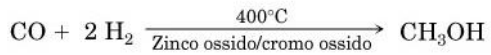


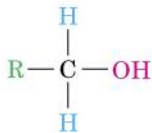
Alcol



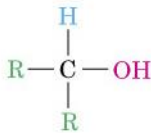
Alcol



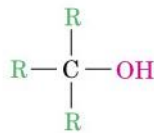
Nomenclatura



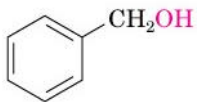
Alcol primario (1°)



Alcol secondario (2°)



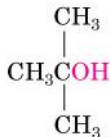
Alcol terziario (3°)



Alcol benzilico
(Fenil metanolo)



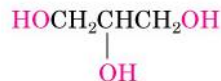
Alcol allilico
(2-Propen-1-olo)



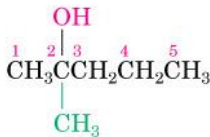
Alcol *tert*-butilico
(2-Metil-2-propanolo)



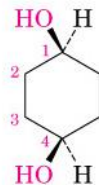
Glicole etilenico
(1,2-Etandiolo)



Glicerolo
(1,2,3-Propantriolo)



2-Metil-2-pentanolo

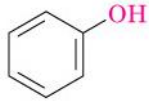


cis-1,4-Cicloesandiolo

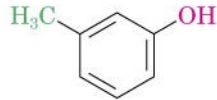


3-Fenil-2-butanolo

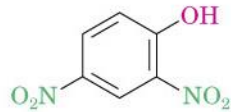
Fenoli



Fenolo



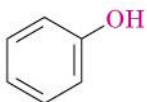
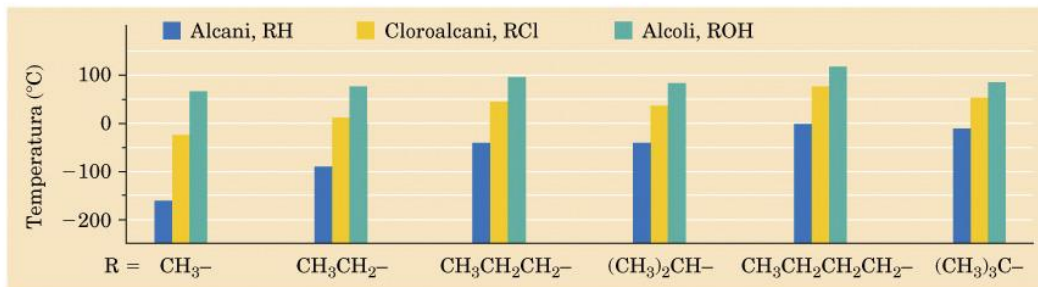
***m*-Metilfenolo
(*m*-Cresolo)**



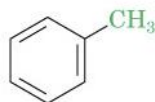
2,4-Dinitrofenolo

Legame a idrogeno

Confronto dei punti di ebollizione di alcuni alcani, cloroalcani ed alcoli. Gli alcoli mostrano generalmente i punti di ebollizione più alti.



Fenolo: p.e. = 181.7°C

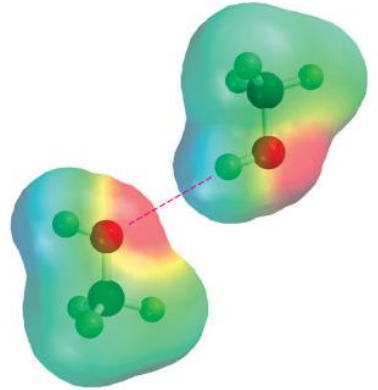
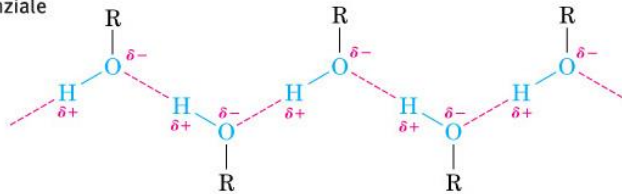


Toluene: p.e. = 110.6°C

Legame a idrogeno

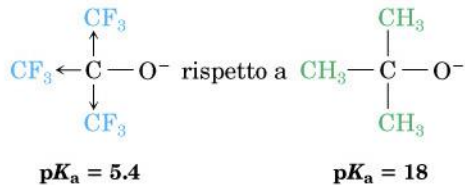
Il legame a idrogeno negli alcoli e nei fenoli. Le molecole sono tenute unite da una debole attrazione che si instaura tra un idrogeno polarizzato positivamente di un gruppo OH e l'ossigeno polarizzato negativamente di un altro gruppo OH.

La mappa di potenziale elettrostatico del metanolo mostra chiaramente l'idrogeno del legame O—H polarizzato positivamente (zona blu) e l'ossigeno polarizzato negativamente (zona rossa).

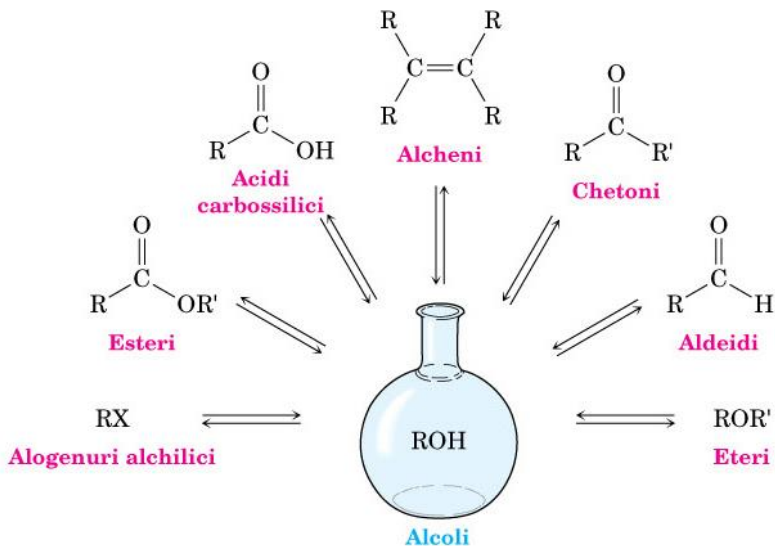


Acidità degli Alcol

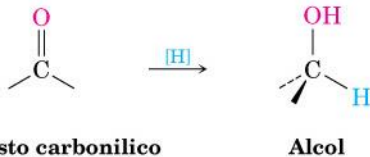
I gruppi elettron-attrattori stabilizzano lo ione alcossido ed abbassano il pK_a



Il ruolo centrale degli alcoli nella chimica organica



Riduzione di composti carbonilici

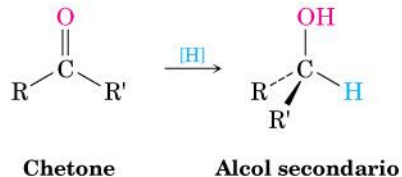
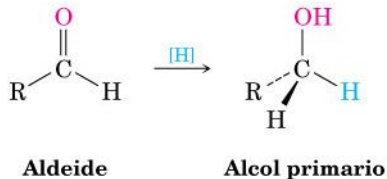
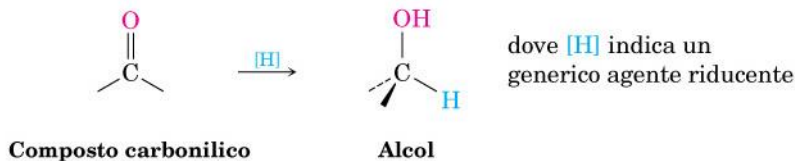


dove $[H]$ indica un generico agente riducente

Composto carbonilico

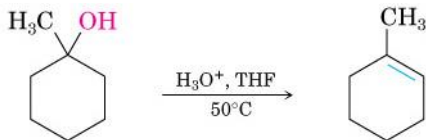
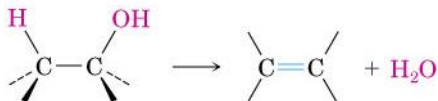
Alcol

Riduzione di composti carbonilici



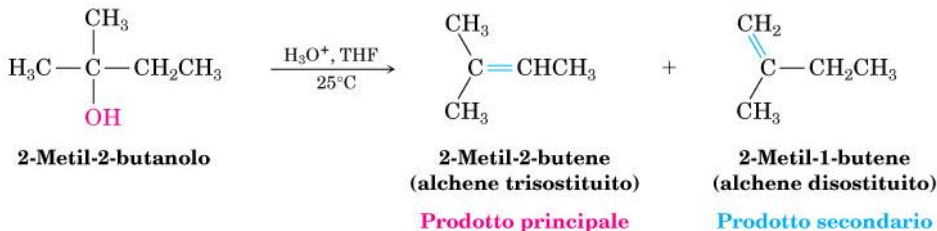
Disidratazione ad alcheni

Reazione di
disidratazione



1-Metilcicloesano

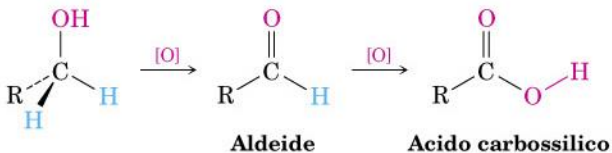
1-Metilcicloesene (91%)



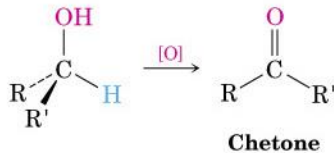
Ossidazione a composti carbonilici



Alcol primario



Alcol secondario



Alcol terziario

