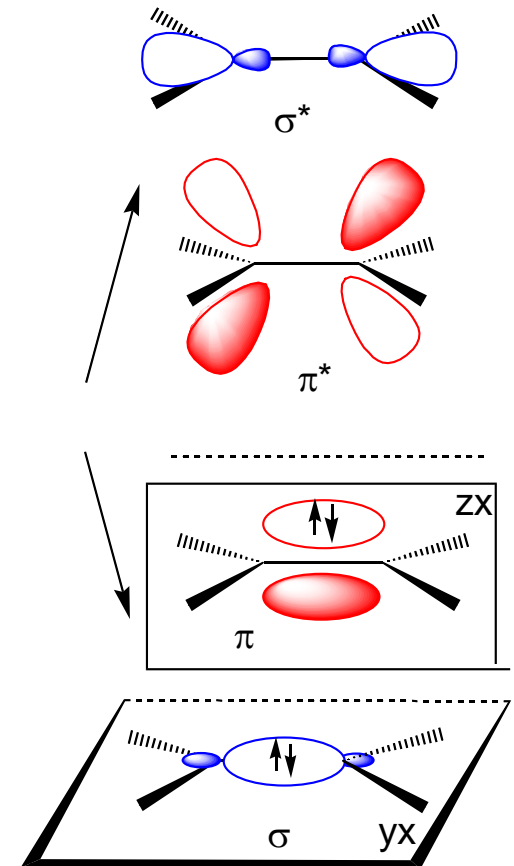
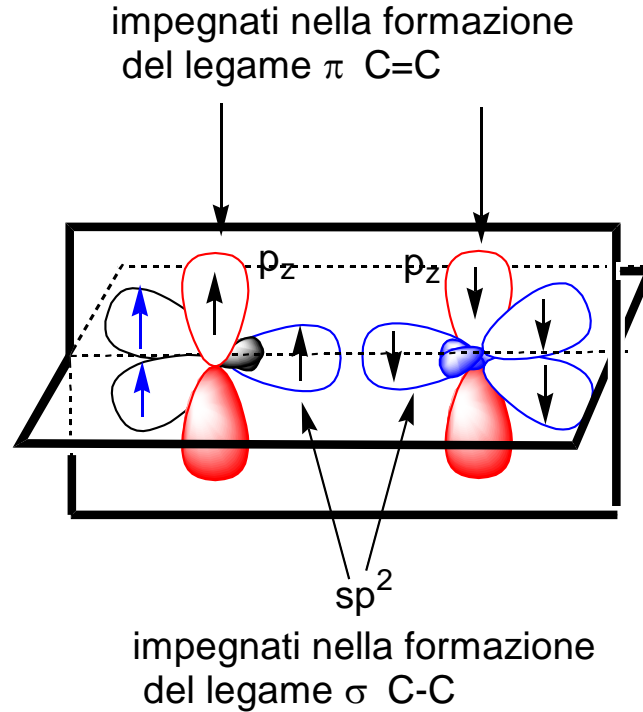
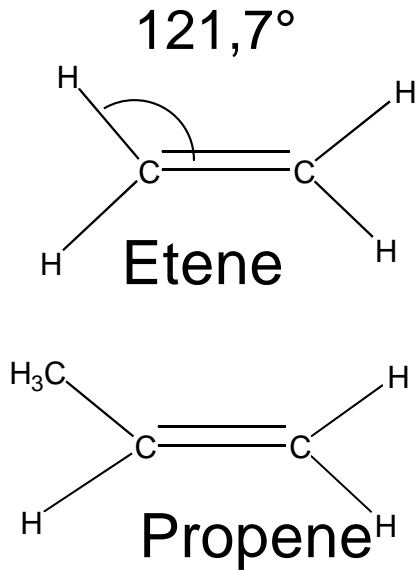


# Alcheni (olefine)

Formula generale  $C_nH_{2n}$

idrocarburi contenenti il doppio legame C=C

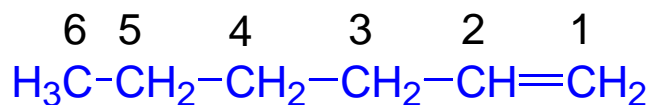
Il carbonio impegnato nel doppio legame è ibridato  $sp^2$



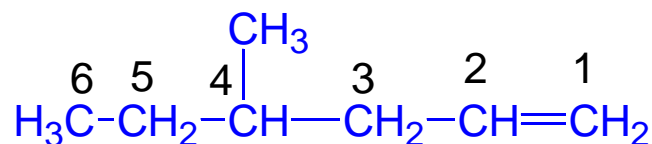
# Nomenclatura degli alcheni

IUPAC

Cambiando il suffisso -ano (degli alcani) in -ene

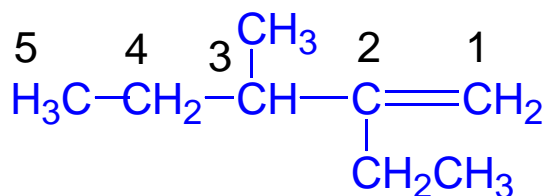


1-esene



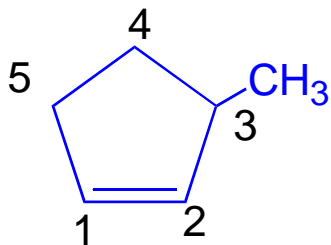
4-metilesene

Si prende la catena più lunga contenente il doppio legame

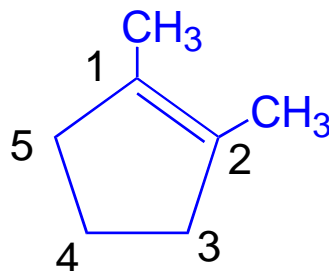


2-etil-3-metilpentene

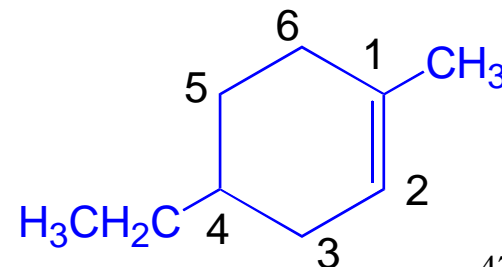
## Cicloalcheni



3-metilciclopentene



1,2-dimetilciclopentene

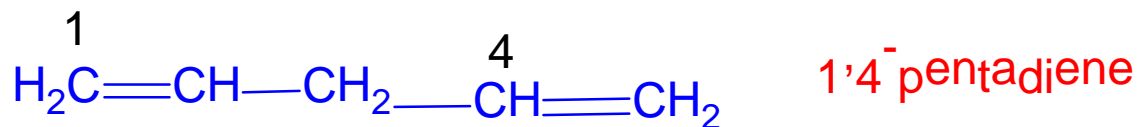


4-etil-1-metilcicloesene

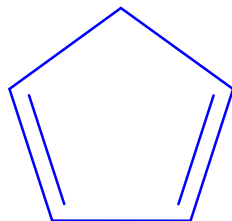
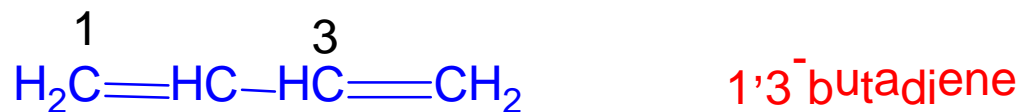
# Dieni

Gli alcheni con due doppi legami sono detti dieni

## Dieni isolati



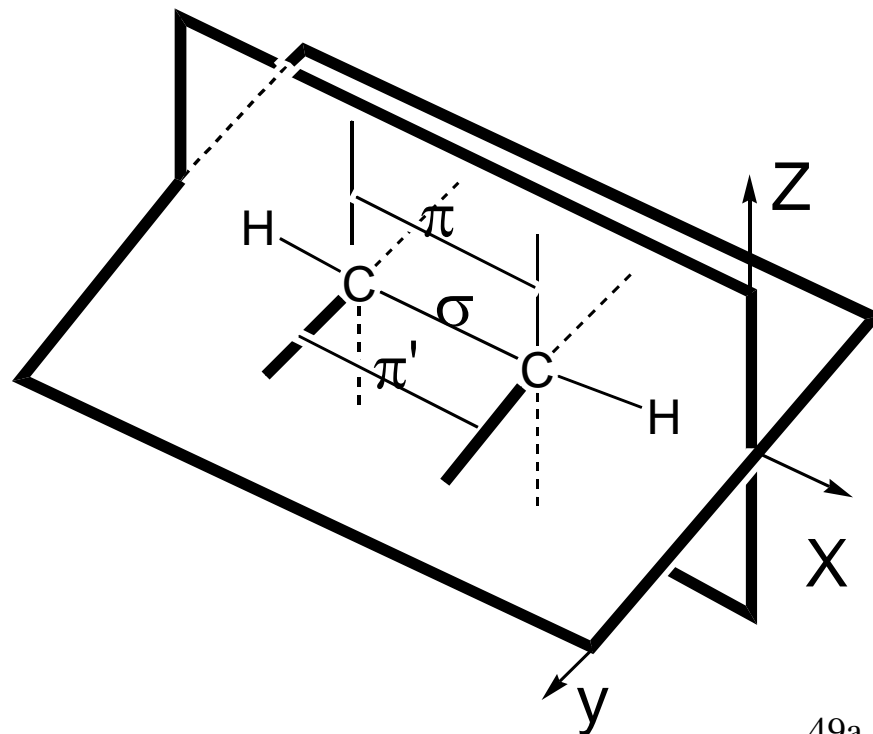
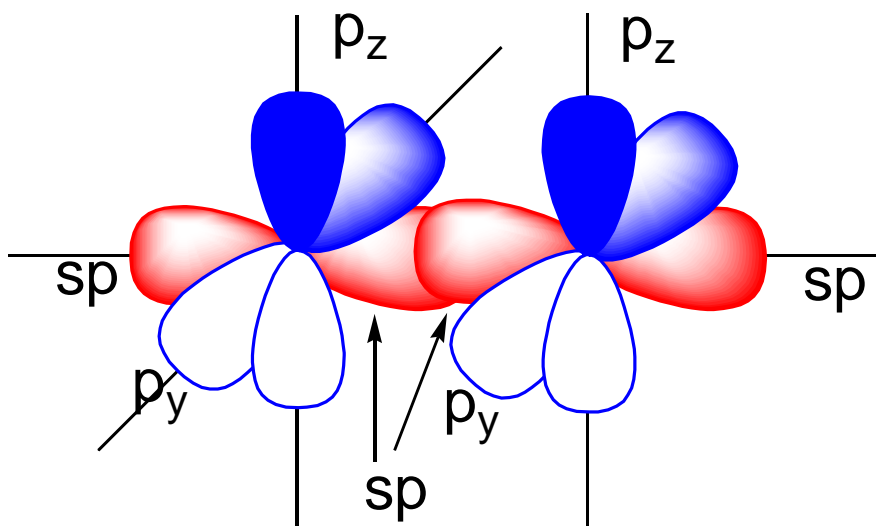
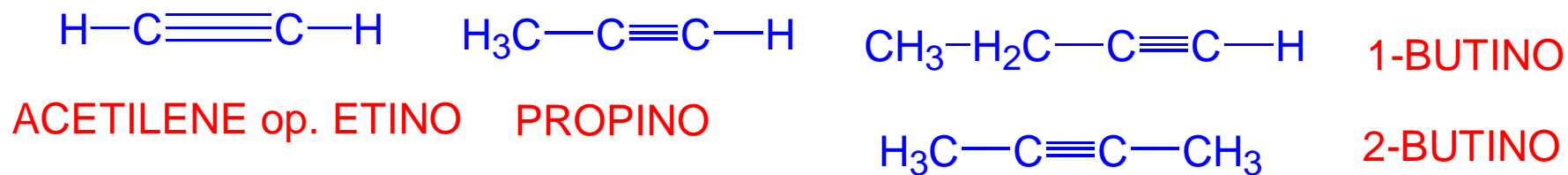
## Dieni coniugati



1,3-ciclopentadiene

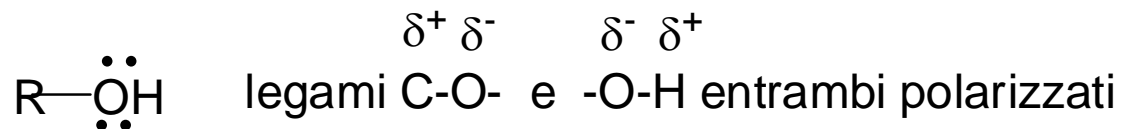
# Alchini $C_nH_{2n-2}$

I carboni del triplo legame sono ibridati sp



# Alcoli

Formalmente derivano da un alcano per sostituzione di un idrogeno con l'ossidrile (-OH), che quindi è legato ad un C sp<sup>3</sup>



CH<sub>3</sub>OH **alcol metilico o metanolo**

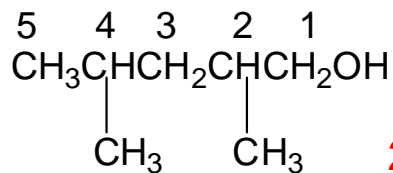
CH<sub>3</sub>CH<sub>2</sub>OH **alcol etilico o etanolo** **alcol 1°**

CH<sub>3</sub>-CH-CH<sub>3</sub>  
|  
OH **alcol isopropilico o 2-propanolo** **alcol 2°**

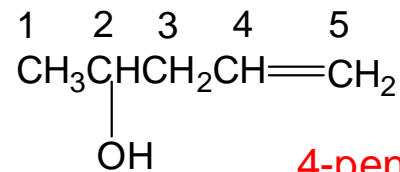
CH<sub>3</sub>  
|  
CH<sub>3</sub>-C-CH<sub>3</sub>  
|  
OH **alcol *terz*-butilico o 2-metil-2-propanolo** **alcol 3°**

# Alcoli

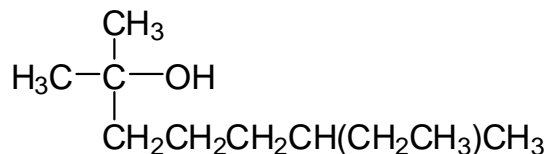
La catena lineare più lunga contenente il carbonio con l'OH fornisce il nome base dell'alcol che deve terminare con la desinenza **-olo**. La numerazione deve essere tale da avere il carbonio con l'OH con il numero d'ordine più basso



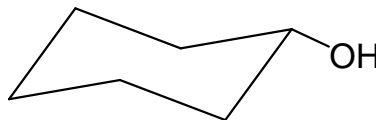
2,4-dimetil-1-pentanol



4-penten-2-olo



2,6-dimetil-2-ottanol



cicloesanol

## Alogenuri alchilici e arilici

Derivano formalmente da un alcano in cui un idrogeno è stato sostituito da un alogeno, il quale quindi è legato ad un C  $sp^3$



**IUPAC** Il nome dell'alcano è preceduto dal nome dell'alogeno

$\text{CH}_3\text{Cl}$  clorometano o cloruro di metile

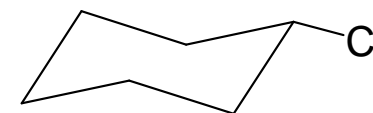
$\text{CH}_3\text{CH}_2\text{Cl}$  cloroetano o cloruro di etile **alogenuro 1°**

$\text{H}_3\text{C}-\underset{\text{Cl}}{\text{CH}}-\text{CH}_3$  2-cloropropano o cloruro di isopropile **alogenuro 2°**

$\text{H}_3\text{C}-\underset{\text{Cl}}{\overset{\text{CH}_3}{\text{C}}}-\text{CH}_3$  2-cloro-2-metilpropano o cloruro di *terz*-butile **alogenuro 3°**

$\text{CHCl}_3$  triclorometano  
o **cloroformio**

$\text{CCl}_4$  tetraclorometano o  
**tetracloruro di carbonio**



clorocicloesano